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PtCl₂- and PtCl₄-Catalyzed Cycloisomerization of Polyunsaturated Precursors

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The PtCl₂- and PtCl₄-catalyzed cycloisomerization of polyunsaturated precursors with different O-protecting groups at the propargylic position is reported. The free hydroxy, O-(bromomethyl)dimethylsilyl, O-methyl and O-propenyl derivatives 1a-f have afforded mixtures of skeletal rearrangement products and/or polycyclic molecules incorporating a cyclopropyl ring system in variable chemical yields. Other parameters such as the variation of the alkenyl chains and the triple-bond substitution have been examined and gave new mechanistic elements. Very interestingly, for analogous compounds containing an ester moiety, such as the O-4-nitrobenzoyl or acetate group (1g-j, and 22) bicyclo[4.1.0]hept-2enyl derivatives were obtained in good yields. These products are formally the result of a 1,2-migration of an O-acyl group, coupled to a cyclopropanation step. Related propargylic acetates with a pendant aromatic group (25) or substituted at the terminal acetylenic carbon with a methoxycarbonyl group (24) did not afford this class of rearranged compounds. This type of reaction seems restricted to propargylic derivatives, because the suitably functionalized homopropargylic acetate 24 only afforded 1,3-dienes.

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Introduction

In the last few years, our groups have been actively working on the Pauson-Khand reaction^[1] and free radical chemistry^[2] of diversely functionalized 1,6-enynes, and in this context, we have recently investigated the reactivity of substrates of type 1 (Figure 1).

Figure 1. General 1,6-enyne framework.

Continuing with our interest in the straightforward assembly of complex polycyclic structures from readily available precursors, we have concentrated on the PtCl2-catalyzed cycloisomerization of these polyunsaturated precursors. In fact, transition-metal-catalyzed cycloisomerization of 1,6-enynes^[3,4] constitutes one of the best-known examples of what have been called *full atom economy reactions*.^[5] Since the pioneering work of Trost with electrophilic palladium catalysts, [6] some novel reactions featuring skeletal rearrangements, [7] and cyclopropanation processes have been reported. [8] With PtCl₂, [9,10] or Au^I and Au^{III} catalysis, [11-14] and very recently Cu^I, [15,16] FeCl₃[17] and In^{III}, [18,19] a number of reports from different laboratories have highlighted the scope and limitations of this new synthetic opportunity for the preparation of carbocycles and/or heterocycles.

Very pleasantly, the PtCl₂-mediated cycloisomerization of the substrates 1 (Figure 1) gave new and surprising results.[20] Subsequent chemical developments of these results,[21-23] as well as our theoretical contributions to this area have also been recently reported. [24-26] Now, in this paper, we would like to describe in full some of the preliminary results^[20] of the PtCl₂-promoted cycloisomerization of precursors 1, as well as new substrates for this reaction, which illustrate the synthetic power of these transformations.

Results and Discussion

1. Preliminary Results

The choice of substrates 1 relied on several features: an easily accessible branched structure, with two pendant ole-

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fins connected by distinct tethers and a functionalized, easily modified and stereogenic propargylic position. When submitted to typical reaction conditions (5 mol-% PtCl₂, toluene, 80 °C for 2.5 h), precursor 1a provided a mixture of the expected formal metathesis adduct 2a as a minor fraction and a larger amount of a fairly complex structure, compound 3a (48% yield, Scheme 1).

Inspection of the NMR spectrum of **3a** clearly showed us that no olefinic protons were present, but the high density of protons at high field suggested that these molecules were possibly polycyclic derivatives incorporating one or more cyclopropyl ring systems. Several structural hypotheses were raised, but a definitive structure was impossible to propose at this point. Finally, the solution to this problem was obtained by X-ray diffraction analysis^[20] of the 4-nitrobenzoyl derivative of compound **3a** (**4**), easily prepared by the standard method in an unoptimized reaction (Scheme 2).

The structure of compound **3a** (Scheme 1) corresponds to the octahydro-2,2-dimethyl-1*H*-dicyclopropa[*a*,*g*]pentalene skeleton, which had not been described previously in the literature. Moreover, the formation of compound **3a** that involves the formation of four C–C and four new stereocontrolled centers, is spectacular in terms of chemo-, re-

gio- and stereoselectivity, featuring a synthetically simple and efficient operation. Additionally, this reaction involves the transformation of the alkyne moiety into a bis-carbene entity that accomplishes a bis-cyclopropanation, leading to two fused bicyclo[3.0.1]hexane systems. Such bis-cyclopropanation had been previously found by Chatani and Murai with a linear dienyne.^[27] Our branched system leads to a particularly strained structure, which illustrates the synthetic power of these reactions. Recently, gold^[28] and ruthenium^[29] catalyses have also given interesting entries into bis-cyclopropyl derivatives.

The observed spectroscopic data of compounds **3b** and **3c**, similar to that of the analogous derivative **3a**, prompted us to assign the same structure to these molecules, as well as the same relative configuration at the newly formed stereocenters. As an additional proof of structure, the reaction of silyl derivative **3b** with tetrabutylammonium fluoride in THF at room temperature afforded a compound identical to polycycle **3a** in quantitative yield (Scheme 1). Interestingly, methyl ether **1c** gave diastereomerically pure **3c**, in a very clean reaction (73% yield), with no detectable metathesis adduct **2c**. The use of PtCl₄ also proved to be very rewarding for this reaction, since tetracycle **3c** was isolated in 77% yield. In contrast, methallyl precursor **1d** underwent

$$\begin{array}{c|c}
R & OX \\
R & PtCl_n (5 \text{ mol-}\%) \\
\hline
PhMe, 80 °C
\end{array}$$

$$\begin{array}{c} R \\ R \\ \end{array} \begin{array}{c} OX \\ R' \\ \end{array} \begin{array}{c} + R \\ R \\ \end{array} \begin{array}{c} OX \\ \\ H \\ \end{array} \begin{array}{c} R' \\ \\ H \end{array}$$

Entry	X	R,R'	Catalyst	2	3
1, 1a	Н	Me, H	PtCl ₂	2a, 20%	3a, 48%
2, 1b	SiMe ₂ Br	Me, H	$PtCl_2$	2b, 8%	3b, 38%
3, 1c	Me	Me, H	$PtCl_2$	-	3c, 73%
4, 1c	Me	Me, H	$PtCl_{4}$	_	3c, 77%
5, 1d	Me	Me, Me	PtCl ₂	-	3d , 34%
6, 1 d	Me	Me, Me	$PtCl_4$	-	3d, 40%
7, 1e	Me	H, H	PtCl ₂	-	3e, 54%
8, 1e	Me	H, H	PtCl ₄	-	3e , 20%

Scheme 1.

OH
$$O_2$$
 O_3 O_4 O_4 O_5 O_7 O_8 O

Scheme 2.

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in a clean reaction from which **3d** was isolated as a single product in 34% yield with PtCl₂ and in a slightly better yield (40%) with PtCl₄ (Scheme 1, entries 5 and 6). We could also show that the presence of a dimethyl group on the butenyl chain was not necessary, since we could isolate **3e** in 54% yield consistently as a single diastereomer from precursor **1e**. In that case, the use of PtCl₄ did not result in any improvement of the yield.

These promising findings prompted us to expand and explore this chemistry by preparing new precursors for the Pt^{II}-mediated cycloisomerization, as well as to define a mechanistic rationale.^[20,24–26] Thus, all these results were consistent with the intervention of a platinacyclo-propyl intermediate^[7–9,29–31] of type I that can evolve into a formal

Scheme 3.

Scheme 4.

metathesis pathway or can be further trapped to give the tetracyclic structures 3a-e (Scheme 3).

Additional information was collected by examining the behavior of *O*-allyl precursor **1f**, which contains an additional unsaturation (Scheme 4) and is easily prepared from precursor **1a**.

Despite all the possible interventions of the unsaturations, a clean reaction (room temp. for 16 h) was observed, since two major adducts (3f and 5) were isolated in good overall yield (44% and 36% yield, respectively, Scheme 4) and in both cases as single diastereoisomers. The structure of product 3f was proposed by analogy with spectroscopic data of the products 3a–e. Structural assignment of product 5 was secured by derivatization and X-ray analysis of 6 (Scheme 5).^[32] These findings are supported by recent mechanistic studies carried out by Soriano and Marco-Contelles^[33] and represent an interesting possibility for forming oxygenated heterocycles.

2. Variations on the *O*-Methyl Substrates

Starting from the general framework 1, we enlarged our study, now considering 1,6-enynes 7, which feature the easily modifiable groups R, R' and R''(Figure 2).

Figure 2. 1,6-Enynes bearing a propargylic OMe group.

Alkenyl Chain (R' Group)

We first looked at the possibility of obtaining tetracycles incorporating medium-sized rings that would result from the trapping of carbene I with a long pendant alkene chain. For this purpose, we prepared dienyne 7a, expecting that it would lead to a bis-cyclopropyl-substituted 5,8-bicyclic

Scheme 5.

Scheme 6.

system. When we submitted this compound to typical conditions, we did not observe any polycyclic compound but a mixture of two products. Metathesis adduct 8 was obtained in 60% yield, along with diene 9 in 16% yield (Scheme 6). To the best of our knowledge, although frequently occurring when other metals are used, [4] this is the first report of the formation of a diene of type 9 in a Pt-catalyzed cycloisomerization. Only the 1,6-enyne moiety was involved in this reaction, most probably because the R' alkene is too remote. Consequently, in contrast to the findings shown in Scheme 1, the formal metathesis product was the major one. We decided to test the reactivity of deuterated compound 7aD for evidence of atom scrambling, as previously described by Murai^[34] and Oi and Inoue^[35] in this kind of cycloisomerization.[36] When this compound was submitted to the same experimental conditions, it led to deuterated

Scheme 7.

Scheme 8.

dienes **8D** in a 7:3 ratio in favor of the abnormal metathesis product. It should be noted that the stereochemistry of the deuterium incorporation was fully consistent with literature reports.^[3,35,37] The stereochemical assignment of *exo* diene **9D** was based on nOe measurements (Scheme 6).

We then prepared a new precursor that possessed a *gem*-dimethyl moiety on the R' chain. We hoped that a Thorpe–Ingold effect would favor the formation of the cyclooctane product.^[38] We were confident, from the results obtained with **7a**, that the first cyclopropanation would take place easily, generating the platinacyclopropyl carbene species. When we submitted **7b** to cycloisomerization conditions, we did not obtain the expected product but a mixture of two compounds, cyclohexene **10** in 45% yield and cyclopentane **11** in 25% yield, analogous to **9**. No trace of the metathesis product was observed (Scheme 7).

Eventually, we supressed the pendant allylic chain to focus on the 1,6-enyne reactivity. Substrate **7c** reacted sluggishly with PtCl₂ and even worse with PtCl₄. Addition of an olefin^[39,40] to the reaction medium allowed for the formation of metathesis adduct **12** as the major compound, along with diene **13** in a moderate overall yield.

Substrate **7d**, which does not bear the *gem*-dimethyl moiety, gave a complex mixture from which no metathesis product was observed and **14** and **15** could be isolated, albeit in low yield (Scheme 8).

Triple Bond Substitution (R'' Group)

To gain more insight into this highly substrate-dependent reaction, we also examined the substitution of the triple bond. The easily synthesized compound 7e, which that features a TMS group on the alkyne, showed no reaction towards PtCl₂ or PtCl₄ catalysis. This lack of reactivity has already been described with other silylated systems^[23,41] (Scheme 9). In the same vein, alkyl substitution gave a very low yield of cycloisomerization products (<15%). So, we turned our attention to precursors bearing an electron-withdrawing group. Methoxyester derivative 7f reacted readily with a good overall yield. Major product 16 came from a formal metathesis reaction, and minor product 17

was the dimethylenecyclopentane. No reaction was observed with the pendant allylic moiety (Scheme 9).

All this led us to refine our mechanistic proposal. All of our findings and those reported in the literature urged us to consider cyclopropylplatinacarbene I as the key intermediate in these transformations. Trapping of this species by a pendant unsaturation would lead to polycycles 3.

Other evolutions are possible. Metallacarbene I can also be viewed as a metallic zwitterion I', which corresponds to a metalated non-classical homoallylic cation.^[9,31] Zwitterionic cyclohexyl intermediate II would contribute significantly only when R = H, which alleviates severe 1,3-diaxial interactions. Consecutive proton eliminations, possibly assisted by platinum, affords methylenecyclohexene adducts 10 and 14. The formation of dienes 9, 11, 13, 15, and 17 (and notably 9D) is a signature of platinacyclopentene III. This intermediate can originate from cyclopropylplatinacarbene I through a vinylcyclopropane to cyclopentene type of mechanism or by direct oxidative cyclization after coordination of both unsaturations of the 1,6-enyne moiety. However, this possibility has not been supported by DFT calculations, [30] and the observation of products deriving from this intermediate is generally scarce. Therefore, we favor the intervention of platinacyclopentene III as an evolution of cyclopropylplatinacarbene I. β-hydride elimination gives birth to intermediate IV, which leads to dimethylenecyclopentane adducts after reductive elimination.

The formation of formal metathesis adducts **8**, **12**, and **16** from intermediate **I** has been fully studied recently. Calculations by Soriano and Marco-Contelles on PtCl₂ and by Echavarren on gold complexes corroborate the initial mechanism proposal by Murai, Inoue and Oi. [25,34–36] Single cleavage or double cleavage of the cyclopropyl moiety explain the formation of both isomers.

If the triple bond is substituted by a TMS group or an alkyl chain, unproductive and unselective catalyses are observed. In contrast, substitution by an electron-withdrawing group (precursor 7f) restores a decent level of reactivity, as evidenced by the formation of dienes 16 and 17, but does not allow the formation of tetracyclic adducts, probably due to unfavorable steric interactions.

OMe
$$\frac{\text{PtCl}_2 \text{ or PtCl}_4, 5 \text{ mol-}\%}{\text{toluene, } 80 \text{ °C, } 1\text{d, then } \Delta, 2\text{ d}} \text{ no reaction, starting material}$$

$$7e$$

$$\frac{\text{OMe}}{\text{PtCl}_2 \text{ (5 mol-}\%)}$$

$$\frac{\text{PtCl}_2 \text{ (5 mol-}\%)}{\text{PhMe, } 80 \text{ °C, } 1\text{ h}}$$

$$CO_2\text{Me}$$

$$7f$$

$$16, 46\%$$

$$17, 18\%$$

Scheme 9.

Scheme 10.

Finally, an interesting observation stems from the difference of reactivity between **7a** and **7c** and **7b** and **7d**, the latter giving more sluggish results. To account for this, we suspect that an additional coordination of the metal by the remote unsaturation would somehow stabilize intermediate I and favor its evolution through other pathways (e.g. diene formation) when the second cyclization is too difficult to take place. The same process would take place in the presence of an additional olefin.

All these considerations are summarized in Scheme 10.

In conclusion, we have carried out several experiments on 1,6-enynes possessing a methoxy moiety on the propargylic position, making subtle variations on the general framework. The study of the influence of these parameters (gem-dimethyl on the enyne, additional unsaturation, triple bond substitution) allowed us to draw an overview of the different mechanistic pathways involved from a single common carbene intermediate. DFT calculations to support these conclusions are ongoing and will be reported in due course.

3. Variations on the *O*-Acyl Substrates

Next, we considered precursors **1g**–**i**, where a 4-nitroben-zoyloxy group or an acetate has been located at the propargylic position. These compounds were easily synthesized by standard *O*-acylation (Et₃N, DMAP in CH₂Cl₂) of the cor-

responding free alcohols. In fact, for correlation purposes, we wanted to prepare product 4 (Scheme 2), which we had previously obtained by acylation of polycycle 3a, by direct PtCl₂-cycloisomerization from the corresponding open precursor 1g. In Scheme 11, we show the results we observed in this PtCl₂-catalyzed cycloisomerization reaction. Contrary to our expectations, esters 1g, 1h and 1i gave the bicyclo[4.1.0]hept-2-en-2-yl 4-O-acyl derivatives 18a (76%), 18b (88%) and 18c (62%), respectively, in good yields and in shorter reaction times than those of the O-alkyl derivatives. These compounds were accompanied by minor amounts of cyclopentene derivatives 19a, 19b and 19c, whose structure assignment was based on ¹H NMR spectra. No pure sample of 19 could be obtained, as it was always in a mixture with the corresponding product 18. Unmethylated precursor 1j could also efficiently undergo the cycloisomerization process to provide 18d with PtCl₂ or PtCl₄. In that case, no five-membered ring was detected. Additional proof of structure was obtained when compound 18b was transformed into trans ketone 20 after methanolysis (Scheme 11). The stereochemical assignment of 20 was deduced from nOe studies.

Interestingly, a careful re-examination of the current literature showed that this PtCl₂-mediated cycloisomerization reaction of propargyl acetates was not entirely new, as some precedent was established by Ohloff in 1976 with Zn^{II}. [42] Some years later, Rautenstrauch also described the same

Entry	X	R,R'	Catalyst	18	19
1, 1g 2, 1h 3, 1i 4, 1j 5, 1j	p-COC ₆ H ₄ NO ₂ COCH ₃ p-COC ₆ H ₄ NO ₂ COCH ₃ COCH ₃	Me, H Me, H Me, Me H,H H,H	PtCl ₂ PtCl ₂ PtCl ₂ PtCl ₂ PtCl ₄	18a, 76% 18b, 88% 18c, 62% 18d, 93% 18d, 98%	19a, 4% 19b, 3% 19c, 16%

Scheme 11.

reaction,^[43] and found that similar products were obtained in moderate yields (10–40%) from related precursors, with PdCl₂(CH₃CN)₂ as a catalyst,. After this preliminary communication, no further reports from this laboratory were available, and the potential interest of this new mode of cycloisomerization reaction, as well as the potential synthetic applications of the resulting building blocks, remained unexplored. Recently, after we recently rediscovered this reactivity,^[20] Ohe and Uemura have reported the intermolecular version^[44,45] of this reaction. Further reports from our laboratories,^[46,47] as well as the development of gold(I) or gold(III) catalyses,^[48,49] have established the versatility of this transformation and have enlarged its scope to even include asymmetric transformations.^[15,50–52]

Interestingly, a *gem*-dimethyl effect^[38] was observed in the case of **21**, which preferably gave **22** over **23** (Scheme 12).

Scheme 12.

In summary, we have observed the important effect that the type of the substituents at propargylic centers in differently functionalized 1,6-enynes have on the course of their PtCl₂-catalyzed cycloisomerization reactions. It was noticed that on going from the free alcohol (or ethers) to *O*-acyl derivatives, the major final resulting products were completely different. For the *O*-acyl migration, the mechanism may involve an initial cycloisomerization followed by a migration, with the initial 1,2-migration being followed by intramolecular cyclopropanation or cyclization of a half-rearranged intermediate.^[15,50–51] DFT calculations on our substrates have not allowed us to discern an obvious pathway so far.^[24] This result urged us to extend these observations to other enynes, in order to explore the scope and generality of the process. We could notably show that the substitution on the alkyne favors the formation of allenyl esters that can further undergo a thermal [3,3] rearrangement.^[23]

We have also prepared and submitted to cyclization precursors 24, 25 and 26 (Figure 3). In compound 24, we have incorporated a propargylic acetate and a methoxycarbonyl group at the terminal position of the acetylenic moiety. Propargylic acetate 25 has been designed to determine if the presumed Pt carbene intermediate could be trapped by the pendant aromatic ring. Finally, we wanted to know if this cycloisomerization, as shown in Scheme 8, is still possible with the homopropargylic acetate 26 or is it limited to propargylic carboxylates.

$$OAc$$
 OAc
 OAC

Figure 3. Different enynes bearing an OAc group.

Precursor **24** was prepared from commercial 5-hexene-2-one **(27)**, via alcohol **28**, as shown in Scheme 13. Under the usual catalysis conditions, we only isolated the acyclic

product **29** in low yield (11%). The structure of this compound was firmly established based on its spectroscopic and analytical data. The assigned configuration at the double bonds rests on selective nOe Experiments. As observed before,^[23] the formation of compound **29** could simply be explained by hydrogen abstraction by Pt to form a dienylplatinum hydride that undergoes reductive elimination. The reasons why the expected product from a Rautenstrauch isomerization^[20,43] have not been detected in this case are still obscure and demand further investigation. In fact, the reaction of terminal substituted alkynes seems to be very substrate-dependent, as recent results have shown that conveniently functionalized 5-en-1-yn-3-ol derivatives substituted with methoxycarbonyl^[21] groups at the terminal alkyne moiety efficiently cyclize to give bicyclic adducts.

a: methyl propiolate, LDA, THF, -78 °C (71%) b: Ac₂O, py, DMAP, r.t., 2.5 h (73%) c: PtCl₂ (5 mol-%), toluene, 80 °C, 11 h

Scheme 13.

We have prepared precursor **25**^[53] from commercially available 4-phenyl-2-butanone (**30**), via propargylic alcohol **31** using standard protocols (Scheme 14). After PtCl₂-mediated cycloisomerization, we were able to isolate and characterize only the 1-acetoxyallene **32** that would originate from a 1,3-acetate migration (Scheme 14). Formal C–H activation products consisting of an indene motif have been obtained with ruthenium, [44] platinum [54] and gold catalysis [55] when the aromatic group is at the propargylic substitution.

Similarly, we have synthesized compound **26** from 6-methylhept-6-ene-2-one (**33**), via known alcohol **34**.^[56] The PtCl₂-promoted cycloisomerization of precursor **26** gave an inseparable mixture of products **35** and **36** (Scheme 15) in good yield. The structure of these molecules was clearly established after careful spectroscopic analysis. Major isomer **35** is the skeletal rearrangement product,^[12] while compound **36** is a typical 1,3-diene obtained in transition-metal promoted reactions.^[4] In no case was a cyclopropyl-derived

a: ethynylmagnesium bromide, r.t., 12 h (86%)

b: Ac₂O, py, DMAP, r.t., 12 h (70%)

c: PtCl₂ (5 mol-%), toluene, r.t., 5 h

Scheme 14.

product detected and isolated, implying that the Rautenstrauch isomerization^[43] seems restricted to propargylic carboxylates.

26

a: propargyl bromide, Zn, r.t., 19 h (12%) b: Ac₂O, py, DMAP, r.t., 16 h (44%) c: PtCl₂ (5 mol-%), toluene, 40 °C, 40 h

Scheme 15.

Conclusions

In conclusion, the PtCl₂- and PtCl₄-catalyzed cycloisomerization of polyunsaturated precursors with different *O*protecting groups at the propargylic position has been reported. As a result, from free hydroxy, O-(bromomethyl)dimethylsilyl or O-alkyl derivatives, we have isolated mixtures of skeletal rearrangement products and/or polycyclic molecules incorporating a cyclopropyl ring system in variable yields. Although some of these reactions have no preparative impact, they allowed for the isolation of new cyloisomerization products, which strongly support the intermediacy of cyclopropylplatinacarbene and platinacyclopentene intermediates and give new mechanistic insight. Very interestingly, for analogous compounds containing an ester moiety, bicyclo[4.1.0]hept-2-en-yl 4-O-acyl compounds were obtained in good yields. These products are formally the result of a sequential process of cycloisomerization and a 1,2-migration of the benzoyloxy group, the exact order of the events being still under scrutiny. A related propargylic acetate substituted at the terminal acetylenic carbon with a methoxycarbonyl group or with a pendant aromatic group did not afford this class of rearranged compounds. Similarly, this type of reaction seems restricted to propargylic derivatives, because a suitably functionalized homopropargylic acetate only afforded 1,3-dienes.

Experimental Section

General: Reactions were carried out under an anhydrous atmosphere of Ar. Glassware was flame-dried under an argon gas flow prior to use. Anhydrous THF and Et₂O were obtained by distillation over sodium/benzophenone under nitrogen and used immediately. Triethylamine, pyridine and diisopropylamine were dried and then distilled from KOH; toluene and CH2Cl2 were dried and then distilled from CaH₂. n-Butyllithium was purchased as 2.5 M solutions in hexanes and titrated before use. Other reagents were commercially available and used without further purification, unless otherwise indicated. TLC was performed on Silica F254 (Merck) and detection by UV light at 254 nm or by charring with phosphomolybdic-H₂SO₄ reagent, and p-anisaldehyde. Column chromatography was performed on Silica Gel 60 (Merck, 230 mesh). ¹H and ¹³C NMR spectra were recorded at room temperature, either at 200 MHz and 50 MHz with an AC200 Bruker spectrometer, at 250 MHz and 63 MHz with an AC250 Bruker spectrometer, at 300 MHz and 75 MHz with a Bruker Avance-300 spectrometer, or at 400 MHz and 100 MHz, respectively, with an ARX400 AVANCE Bruker spectrometer. Chemical shifts are given in ppm and referenced to the residual solvent signal ($\delta = 7.26$ ppm for CDCl₃) for ¹H NMR spectroscopy. For ¹³C NMR, shifts are referenced to the central solvent peak ($\delta = 77.3$ ppm for CDCl₃). Coupling constants (J) are given in Hertz (Hz). The letters m, s, d, t, q, and hept stand for multiplet, singlet, doublet, triplet, quadruplet, and heptuplet, respectively. IR spectra were recorded with a Tensor 27 (ATR diamond) Bruker spectrometer. IR data is reported as characteristic bands (cm⁻¹) in their maximal intensity. The melting points were measured with an SMP3 Stuart Scientific melting point apparatus or with a digital melting point apparatus (Electrothermal) and are uncorrected. Elemental analysis was performed by the Service Régional de Microanalyse de l'Université Pierre et Marie Curie.

Synthesis of the Precursors: The synthesis of **1a** and **1b** and their spectroscopic data have already been described. [1] *O*-Methylation of the tertiary alcohols with NaH and MeI provided **1c** and **1d** in good yields (>80%). Precursor **1e** was prepared by a similar route,

which was also applied to obtain precursors 7.^[1] Similarly, O-allylation of 1a with NaH and allyl bromide gave 1f. O-Acyl precursors 1g–j and 22 were prepared from the corresponding alcohol with acetic anhydride, or p-nitrobenzoyl chloride in pyridine with DMAP, or in CH_2Cl_2 with Et_3N and DMAP.

4-Ethynyl-4-methoxy-6,6-dimethylocta-1,7-diene (1c): IR (neat): $\tilde{\mathbf{v}} = 3300$, 3100, 2950, 2350, 1650, 1460, 1390 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): $\delta = 5.99$ (dd, J = 17.7, 10.9 Hz, 1 H), 5.82 (m, 1 H), 5.12–4.80 (m, 4 H), 3.31 (s, 3 H, OMe), 2.53 (s, 1 H), 2.44 (d, J = 7.4 Hz, 2 H), 1.71 (s, 2 H), 1.18 (s, 3 H, CH₃), 1.12 (s, 3 H, CH₃) ppm. ¹³C NMR (50 MHz, CDCl₃): $\delta = 149.3$ (CH), 133.4 (CH), 118.3 (CH₂, CH=*C*H₂), 109.2 (CH₂, CH=*C*H₂), 84.1 (C), 76.7 (CH), 75.3 (C), 51.1 (CH₃, OMe), 49.6 (CH₂), 44.6 (CH₂), 36.7 (C), 28.5 (CH₃), 28.2 (CH₃) ppm. C₁₃H₂₀O (192.30): calcd. C 81.20, H 10.48; found C 80.91, H 10.72.

4-Ethynyl-4-methoxy-2,6,6-trimethylocta-1,7-diene (1d): IR (neat): \tilde{v} = 3300, 3050, 2950, 2320, 1630, 1450, 1380 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): δ = 6.00 (dd, J = 17.7, 10.3 Hz, 1 H), 4.92–4.77 (m, 4 H), 3.31 (s, 3 H, OMe), 2.53 (s, 1 H), 2.39 (s, 2 H), 1.79 (s, 3 H), 1.73 (s, 2 H), 1.17 (s, 3 H, CH₃), 1.12 (s, 3 H, CH₃) ppm. ¹³C NMR (50 MHz, CDCl₃): δ = 149.5 (CH), 141.6 (C), 115.3 (CH₂, CH=*C*H₂), 109.0 (CH₂, CH=*C*H₂), 84.4 (C), 76.9 (CH), 75.4 (C), 51.1 (CH₃, OMe), 49.7 (CH₂), 47.9 (CH₂), 36.6 (C), 28.2 (2 CH₃), 24.1 (CH₃) ppm.

4-Ethynyl-4-methoxy-3,3-octa-1,7-diene (1e): IR (neat): $\tilde{v}=3309$, 3078, 2922, 2853, 2109, 1642 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta=5.91-5.76$ (m, 2 H), 5.12 (m, 2 H), 4.93 (m, 2 H), 3.35 (s, 3 H, OMe), 2.50 (s, 1 H), 2.47 (d, J=7.0 Hz, 2 H), 2.28–2.13 (m, 2 H), 1.75 (t, J=8.1 Hz, 2 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta=138.5$ (CH), 133.1 (CH), 118.6 (CH₂, CH=*C*H₂), 114.8 (CH₂, CH=*C*H₂), 83.9 (C), 75.9 (C), 75.1 (C), 51.7 (CH₂), 42.5 (CH₂), 37.4 (CH₂), 28.3 (CH₂) ppm.

4-Ethynyl-6,6-dimethyl-4-O-propenyl-1,7-octadien-4-ol (1f): To a cold (0 °C) solution of 4-ethynyl-6,6-dimethyl-1,7-octadien-4-ol (1a, 220 mg, 1.24 mmol) in dry THF (10 mL), NaH (99 mg, 2.48 mmol) was added, and the reaction mixture was stirred at the same temperature for 10 min. Allyl bromide (0.15 mL, 2.48 mmol) and a catalytic amount of tetrabutylammonium iodide were added. The mixture was stirred for 18 h at room temp., until the reaction was complete. The excess of NaH was destroyed by adding AcOH (0.5 mL) and water (10 mL). The aqueous layer was extracted with diethyl ether. The combined organic layers were washed with brine, dried with Na₂SO₄ and concentrated in vacuo. Purification by flash column chromatography (CH₂Cl₂/hexane, 4:96) afforded 1f (202 mg, 81% yield) as a colorless oil. IR (neat): $\tilde{v} = 3304$, 3081, 2985, 1640, 1460, 1381, 1290, 1250 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 5.92$ (dd, J = 17.5, 10.5 Hz, 1 H), 5.80 (m, 2 H), 5.27– 5.18 (m, 2 H), 5.18–4.91 (m, 2 H), 4.72–4.76 (m, 2 H), 4.11 (ddt, J = 12.3, 4.9, 1.7 Hz, 1 H), 3.91 (ddt, J = 12.3, 4.9, 1.7 Hz, 1 H), 2.43 (s, 1 H), 2.38 (m, 2 H), 1.66 (m, 2 H), 1.06 (s, 3 H, CH₃), 1.05 (s, 3 H, CH₃) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 149.8 (CH), 135.6 (CH), 133.8 (CH), 118.6 (CH₂, CH=CH₂), 116.2 (CH₂, CH=CH₂), 109.5 (CH₂, CH=CH₂), 84.8 (C), 76.9 (C), 75.4 (CH), 65.4 (CH₂), 50.5 (CH₂), 45.7 (CH₂), 37.1 (C), 28.9 (CH₃), 28.6 (CH₃) ppm. MS (70 eV): m/z (%) = 218 (1) [M]⁺, 177 (3), 135 (13), 107 (12), 91 (20), 79 (16), 69 (100), 55 (19), 41 (80). C₁₅H₂₂O (218.33): calcd. C 82.52, H 10.16; found C 82.33, H 10.09.

5-Ethynyl-5-methoxy-3,3-dimethylundeca-1,10-diene (7a): IR (neat): $\tilde{\mathbf{v}}=3305,\ 3080,\ 2940,\ 2105,\ 1640\ \mathrm{cm^{-1}}.\ ^1\mathrm{H}\ \mathrm{NMR}\ (400\ \mathrm{MHz},\ \mathrm{CDCl_3}): \delta=6.02\ (\mathrm{dd},\ J=17.3,\ 10.7\ \mathrm{Hz},\ 1\ \mathrm{H}),\ 5.82\ (\mathrm{ddt},\ J=17.1,\ 10.5,\ 6.6\ \mathrm{Hz},\ 1\ \mathrm{H}),\ 5.02\ (\mathrm{dd},\ J=17.3,\ 1.0\ \mathrm{Hz},\ 1\ \mathrm{H}),\ 5.01\ (\mathrm{dd},\ J=17.1,\ 1.5\ \mathrm{Hz},\ 1\ \mathrm{H}),\ 4.96\ (\mathrm{dd},\ J=10.7,\ 1.0\ \mathrm{Hz},\ 1\ \mathrm{H}),\ 4.94\ (\mathrm{dd},\ J=10.7,\ 1.0\ \mathrm{Hz},\ 1\ \mathrm{Hz})$

10.5, 1.5 Hz, 1 H), 3.32 (s, 3 H, OMe), 2.52 (s, 1 H), 2.07 (m, 2 H), 1.76 (m, 2 H), 1.69 (m, 2 H), 1.50–1.30 (m, 4 H), 1.18 (s, 3 H, CH₃), 1.17 (s, 3 H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 149.6 (CH), 139.3 (CH), 114.7 (CH₂, CH=*C*H₂), 109.6 (CH₂, CH=*C*H₂), 85.1 (C), 76.3 (C), 76.2 (CH), 51.2 (CH₃, OMe), 49.7 (CH₂), 40.1 (CH₂), 37.0 (C), 34.1 (CH₂), 29.4 (CH₂), 23.8 (CH₂), 28.8 (CH₃), 28.7 (CH₃) ppm; **7aD** (C₁₆H₂₅DO): ¹H NMR (400 MHz, CDCl₃): the same spectrum as that of **7a**, except no signal at δ = 2.52 ppm. ¹³C NMR (100 MHz, CDCl₃): the same spectrum as that of **7a**, except δ = 85.1 (t, J = 7 Hz) and 76.2 (t, J = 39 Hz) ppm. CIMS: (NH₃) m/z (%) = 253 (15) [M + NH₄]⁺, 236 (21) [M + H]⁺, 221 [M – 15] ⁺, 204 [M – 31] ⁺.

5-Ethynyl-5-methoxy-7,7-dimethylundeca-1,10-diene (7b): IR (neat): $\hat{v}=3300,\ 3070,\ 2940,\ 2100,\ 1640,\ 1460,\ 1075\ cm^{-1}.\ ^1H\ NMR (400\ MHz,\ CDCl_3): <math>\delta=5.81\ (m,\ 2\ H),\ 5.02\ (d,\ J=17.3\ Hz,\ 1\ H),\ 4.98\ (d,\ J=17.7\ Hz,\ 1\ H),\ 4.95\ (d,\ J=11.5\ Hz,\ 1\ H),\ 4.89\ (d,\ J=10.2\ Hz,\ 1\ H),\ 3.29\ (s,\ 3\ H,\ OMe),\ 2.53\ (s,\ 1\ H),\ 2.15\ (m,\ 2\ H),\ 2.02\ (m,\ 2\ H),\ 1.76\ (m,\ 2\ H),\ 1.65\ (d,\ J=14.2\ Hz,\ 1\ H),\ 1.59\ (d,\ J=14.2\ Hz),\ 1.44\ (m,\ 2\ H),\ 1.04\ (s,\ 3\ H),\ 1.03\ (s,\ 3\ H)\ ppm.\ ^{13}C\ NMR\ (100\ MHz,\ CDCl_3): <math>\delta=139.8\ (CH),\ 138.3\ (CH),\ 114.5\ (CH_2,\ CH=CH_2),\ 113.6\ (CH_2,\ CH=CH_2),\ 84.4\ (C),\ 76.2\ (CH),\ 75.5\ (C),\ 50.8\ (CH_3,\ OMe),\ 48.1\ (CH_2),\ 42.9\ (2\ CH_2),\ 39.7\ (CH_2),\ 33.5\ (C),\ 28.7\ (CH_2),\ 28.5\ (CH_3),\ 28.3\ (CH_3)\ ppm.\ C_{16}H_{26}O\ (234.38):\ calcd.\ C\ 81.99,\ H\ 11.18;\ found\ C\ 81.84,\ H\ 11.35.$

5-Methoxy-3,3,5-trimethylhept-1-en-6-yne (7c): IR (neat): $\tilde{v} = 3309$, 3083, 2923, 2854, 2109, 1639 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 6.01$ (dd, J = 17.7, 10.8 Hz, 1 H), 4.92 (dd, J = 17.7, 1.3 Hz, 1 H), 4.87 (dd, J = 10.8, 1.3 Hz, 1 H), 3.33 (s, 3 H, OMe), 2.48 (s, 1 H), 1.83 (d, J = 14.6 Hz, 1 H), 1.76 (d, J = 14.6 Hz, 1 H), 1.42 (s, 3 H), 1.16 (s, 3 H), 1.14 (s, 3 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 149.6$ (CH), 109.7 (CH₂, CH=CH₂), 85.8 (C), 74.9 (CH), 73.4 (C), 52.7 (CH₂), 51.2 (CH₃, OMe), 37.0 (C), 28.9 (CH₃), 28.4 (CH₃), 28.0 (CH₃) ppm.

5-Ethynyl-5-methoxy-non-1-ene (7d): IR (neat): $\tilde{v} = 3310$, 3078, 2926, 2856, 1642 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 5.84$ (ddt, J = 17.4, 10.4, 6.6 Hz, 1 H), 5.05 (dd, J = 17.4, 1.5 Hz, 1 H), 4.96 (dd, J = 10.4, 1.5 Hz, 1 H), 3.33 (s, 3 H, OMe), 2.46 (s, 1 H), 2.17 (m, 2 H), 1.75 (t, J = 8.4 Hz, 2 H), 1.67 (t, J = 7.8 Hz, 2 H), 1.43–1.30 (m, 4 H), 0.92 (t, J = 7.1 Hz, 3 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 138.3$ (CH), 114.5 (CH₂, CH=*C*H₂), 77.2 (CH), 76.1 (C), 74.1 (C), 51.2 (CH₃, OMe), 37.5 (CH₂), 36.9 (CH₂), 28.1 (CH₂), 25.9 (CH₂), 22.9 (CH₂), 14.0 (CH₃) ppm.

(3-Allyl-3-methoxy-hept-6-en-1-ynyl)trimethylsilane (7e): IR (neat): $\tilde{v} = 3078, 2956, 2928, 2856, 2165, 1642 \text{ cm}^{-1}. ^1\text{H NMR (}400 \text{ MHz, CDCl}_3): <math>\delta = 5.83 \text{ (m, 2 H), } 5.10 \text{ (m, 2 H), } 5.05-4.93 \text{ (m, 2 H), } 3.35 \text{ (s, 3 H), } 2.45 \text{ (d, } J = 7.4 \text{ Hz, 2 H), } 2.18 \text{ (m, 2 H), } 1.71 \text{ (t, } J = 8.3 \text{ Hz, 2 H), } 0.18 \text{ (s, 9 H) ppm.} ^{13}\text{C NMR (} 100 \text{ MHz, CDCl}_3): <math>\delta = 138.8 \text{ (CH), } 133.5 \text{ (CH), } 118.4 \text{ (CH}_2, \text{ CH} = \text{CH}_2), \\ 114.8 \text{ (CH}_2, \text{ CH} = \text{CH}_2), \\ 105.8 \text{ (C), } 91.8 \text{ (C), } 76.4 \text{ (C), } 51.8 \text{ (CH}_3, \text{ OMe), } 42.6 \text{ (CH}_2), \\ 37.5 \text{ (CH}_2), \\ 28.6 \text{ (CH}_2), 0.30 \text{ (3 CH}_3, \text{ TMS) ppm.}$

Methyl 4-Allyl-4-methoxy-6,6-dimethyloct-7-en-2-ynoate (7f): To a solution of 1a (384 mg, 2 mmol, 1 equiv.) in dry THF (10 mL) was added nBuLi (0.95 mL of a 2.2 M solution, 2.1 mmol, 1.05 equiv.) at -78 °C. Methyl chloroformate (0.46 mL, 6 mmol, 3 equiv.) was added, and the mixture was allowed to warm to room temp. The reaction was quenched with a saturated NH₄Cl solution and extracted with Et₂O. The combined organic layers were washed with brine, dried with MgSO₄ and concentrated to give 7f (495 mg, 99% yield). IR (neat): \hat{v} = 3081, 2955, 2873, 2230, 1717, 1639 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 5.94 (dd, J = 17.4, 10.6 Hz, 1 H), 5.78 (m, 1 H), 5.11 (m, 2 H), 4.90 (dd, J = 17.4, 1.3 Hz, 1 H), 4.84 (dd, J = 10.6, 1.3 Hz, 1 H), 3.76 (s, 3 H, CO₂Me), 3.33 (s, 3 H,

OMe), 2.47 (m, 2 H), 1.75 (s, 2 H), 1.12 (s, 3 H), 1.10 (s, 3 H) ppm. 13 C NMR (100 MHz, CDCl₃): δ = 153.7 (C), 148.6 (CH), 132.2 (CH), 119.1 (CH₂, CH=*C*H₂), 109.8 (CH₂, CH=*C*H₂), 87.9 (C), 80.3 (C), 75.3 (C), 52.7 (CH₃), 51.5 (CH₃, OMe), 49.5 (CH₂), 43.8 (CH₂), 36.6 (C), 28.6 (CH₃), 27.9 (CH₃) ppm.

1-Allyl-1-ethynyl-3,3-dimethylpent-4-enyl 4'-Nitrobenzoate (1g): IR (neat): $\tilde{v} = 3300$, 3080, 2960, 2115, 1730, 1640, 1610, 1530, 1350 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): $\delta = 8.29$ (d, J = 9.0 Hz, 2 H, Ar), 8.16 (d, J = 9.0 Hz, 2 H, Ar), 6.00 (dd, J = 17.4, 10.6 Hz, 1 H), 5.92 (m, 1 H), 5.18 (d, J = 16.2 Hz, 1 H trans), 5.16 (d, J =11.2 Hz, 1 H cis), 4.95 (dd, J = 17.4, 1.0 Hz, 1 H trans), 4.84 (dd, J = 10.6, 1.0 Hz, 1 H cis), 3.07 (dd, <math>J = 14.0, 7.2 Hz, 1 H), 2.90(dd, J = 14.0, 7.2 Hz, 1 H), 2.76 (s, 1 H), 2.40 (d, J = 15.0 Hz, 1 H)H), 2.11 (d, J = 15.0 Hz, 1 H), 1.23 (s, 3 H, CH₃), 1.20 (s, 3 H, CH₃) ppm. 13 C NMR (50 MHz, CDCl₃): $\delta = 163.0$ (C, C=O), 148.4 (CH), 136.5 (C, Ar), 132.0 (CH), 130.8 (2 CH, Ar), 128.5 (C, Ar), 123.6 (2 CH, Ar), 119.7 (CH₂, CH=CH₂), 110.4 (CH₂, CH=CH₂), 82.8 (C), 79.1 (C), 77.0 (CH), 49.2 (CH₂), 44.8 (CH₂), 36.9 (C), 29.0 (CH₃), 28.0 (CH₃) ppm. GCMS (EI): m/z (%) = 286 (9) [M– C₃H₅], 150 (100). C₁₉H₂₁NO₄ (327.38): calcd. C 69.71, H 6.47, N 4.28; found C 69.65; H, 6.60; N, 4.18.

1-Allyl-1-ethynyl-3,3-dimethylpent-4-enyl Acetate (1h): IR (neat): \tilde{v} = 3300, 3100, 2950, 2350, 1750, 1640, 1440, 1370 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): δ = 5.93 (dd, J = 17.2, 10.3 Hz, 1 H), 5.81 (m, 1 H), 5.13–5.04 (m, 2 H), 4.90 (dd, J = 17.7, 1.5 Hz, 1 H *trans*), 4.86 (dd, J = 10.3, 1.5 Hz, 1 H *cis*), 2.87 (dd, J = 14.3, 7.4 Hz, 1 H), 2.69 (dd, J = 14.3, 7.4 Hz, 1 H), 2.60 (s, 1 H), 2.15 (d, J = 15.3 Hz, 1 H), 1.95 (s, 3 H, OAc), 1.86 (d, J = 15.3 Hz, 1 H), 1.15 (s, 3 H, CH₃), 1.11 (s, 3 H, CH₃) ppm. ¹³C NMR (50 MHz, CDCl₃): δ = 169.4 (C, C=O), 148.7 (CH), 132.3 (CH), 119.1 (CH₂, CH=*C*H₂), 109.7 (CH₂, CH=*C*H₂), 83.5 (C), 77.0 (C), 75.9 (CH), 49.1 (CH₂), 44.6 (CH₂), 36.8 (C), 28.9 (CH₃), 27.7 (CH₃), 22.2 (CH₃, OAc) ppm.

1-Methallyl-1-ethynyl-3,3-dimethylpent-4-enyl 4'-Nitrobenzoate (1i): IR (neat): $\tilde{v}=3300$, 3080, 2960, 2115, 1730, 1640, 1610, 1530, 1350 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): $\delta=8.29$ (d, J=9.2 Hz, 2 H, Ar), 8.18 (d, J=9.2 Hz, 2 H, Ar), 6.02 (dd, J=17.4, 10.6 Hz, 1 H), 4.97 (dd, J=10.6, 1.0 Hz, 1 H cis), 4.86 (d, J=1.2 Hz, 2 H), 4.85 (dd, J=17.4, 1.0 Hz, 1 H cis), 3.06 (d, J=14.4 Hz, 1 H), 2.87 (d, J=14.4 Hz, 1 H), 2.77 (s, 1 H), 2.42 (d, J=15.0 Hz, 1 H), 2.11 (d, J=15.0 Hz, 1 H), 1.82 (s, 3 H, CH₃), 1.24 (s, 3 H, CH₃), 1.20 (s, 3 H, CH₃) ppm. ¹³C NMR (50 MHz, CDCl3): $\delta=163.1$ (C, C=O), 148.5 (CH), 140.2 (C), 136.6 (C, Ar), 130.8 (2 CH+C, Ar), 126.6 (2 CH, Ar), 116.7 (CH₂, CH=CH₂), 110.3 (CH₂, CH=CH₂), 83.1 (C), 79.0 (C), 77.5 (CH), 49.8 (CH₂), 48.2 (CH₂), 36.9 (C), 29.2 (CH₃), 27.9 (CH₃), 24.1 (CH₃) ppm. GCMS (EI): mlz (%) = 286 (5) [M - C₄H₇], 150 (100).

1-Allyl-1-ethynylpent-4-enyl Acetate (1j): IR (neat): \tilde{v} = 3295, 3079, 2980, 2928, 2857, 2118, 1742, 1642 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 5.84–5.76 (m, 2 H), 5.15–4.94 (m, 4 H), 2.79–2.75 (m, 2 H), 2.59 (s, 1 H), 2.23–2.21 (m, 2 H), 2.06–2.04 (m, 1 H), 2.00 (s, 3 H, OAc), 1.91–1.82 (m, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 169.5 (C, C=O), 137.8 (CH), 132.1 (CH), 119.5 (CH₂, CH=*C*H₂), 115.3 (CH₂, CH=*C*H₂), 82.8 (C), 78.2 (C), 75.0 (CH), 42.8 (CH₂), 37.7 (CH₂), 28.5 (CH₂), 22.1 (CH₃, OAc) ppm.

1-But-3-enyl-1-ethynyl-3,3-dimethylpent-4-enyl Acetate **(21):** IR (neat): $\tilde{v} = 3306$, 3081, 2965, 2931, 2873, 2117, 1743, 1642 cm⁻¹.

¹H NMR (400 MHz, CDCl₃): $\delta = 5.94$ (dd, J = 17.4, 10.6 Hz, 1 H), 5.78 (m, 1 H), 5.04–4.86 (m, 4 H), 2.61 (s, 1 H), 2.23–2.17 (m, 4 H), 2.03 (m, 1 H), 1.99 (s, 3 H, OAc), 1.93–1.90 (m, 1 H), 1.16 (s, 3 H), 1.13 (s, 3 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 169.5$ (C, C=O), 137.8 (CH), 132.1 (CH), 119.5 (CH₂, CH=*C*H₂),

115.3 (CH₂, CH=*C*H₂), 82.8 (C), 78.2 (C), 75.0 (CH), 42.8 (CH₂), 37.7 (CH₂), 28.5 (CH₂), 22.1 (CH₃, OAc) ppm.

Methyl 4-Acetyloxy-4-methyloct-7-en-2-vnoate (24): To a solution of disopropylamine (1.07 mL, 7.65 mmol) in THF (40 mL), cooled to -78 °C, nBuLi (4.78 mL, 7.65 mmol, 1.6 m in hexane) was added. The solution was warmed to 0 °C for 15 min. Then, to this solution, cooled to -78 °C, methyl propiolate (0.82 mL, 9.18 mmol) was slowly added. After stirring at -78 °C for 1 h, the reaction mixture was treated with 5-hexene-2-one (27, 0.5 g, 5.1 mmol), stirred at -78 °C for 30 min, and then warmed to room temp. The reaction mixture was diluted with diethyl ether (40 mL), washed with an aqueous solution of sodium bisulfate (1 N, 2×40 mL), dried with Na₂SO₄ and concentrated in vacuo. Flash chromatography, eluting with EtOAc/hexane (5:95), afforded 28 (660 mg, 71% yield) as a yellow oil. IR (neat): $\tilde{v} = 3079$, 2981, 2953, 2235, 1719, 1642, 1592, 1436, 1372, 1256 cm⁻¹. 1 H NMR (300 MHz, CDCl₃): δ = 5.73 (ddt, J = 16.7, 10.1, 6.4 Hz, 1 H), 5.03 (dm, J = 16.7 Hz, 1 H), 4.85 (dm, J = 10.1 Hz, 1 H), 3.67 (s, 3 H, CO₂CH₃), 2.25–2.07 (m, 2 H), 1.77-1.66 (m, 2 H), 1.43 (s, 3 H, CH₃) ppm. ¹³C NMR (75 MHz, CDCl₃): $\delta = 154.3$ (C=O), 138.1, 115.7, 90.9, 75.3, 68.0, 53.2 (CO₂CH₃), 42.2, 29.5, 29.2 ppm. To a solution of compound **28** (400 mg, 2.19 mmol) in dry pyridine (5 mL), DMAP (53.4 mg, 0.44 mmol) and acetic anhydride (336.3 mg, 3.30 mmol) were added. The reaction mixture was stirred for 2.5 h at room temp. The solvent was evaporated, and the crude material was purified by flash chromatography (EtOAc/hexane, 4:96) to furnish 24 (360 mg, 73% yield) as a colorless oil. IR (neat): $\tilde{v} = 3080$, 2981, 2239, 1747, 1720, 1642, 1435, 1370, 1313, 1262 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 5.68 (ddt, J = 16.7, 10.1, 6.4 Hz, 1 H), 4.99 (dm, J = 16.7 Hz, 1 H), 4.85 (dm, J = 10.1 Hz, 1 H), 3.65 (s, 3 H)CO₂CH₃), 2.17–2.00 (m, 2 H), 1.93 (s, 3 H, OCOCH₃), 1.91–1.73 (m, 2 H), 1.59 (s, 3 H, CH₃) ppm. 13 C NMR (75 MHz, CDCl₃): δ = 169.4 (OCOCH₃), 154.1 (C=O), 137.4, 115.7, 87.2, 77.3, 73.9, 53.2 (CO₂CH₃), 40.3, 28.6, 26.1 (CH₃), 21.9 (OCOCH₃). MS $(70 \text{ eV}) \ m/z \ 209 \ (1) \ [M-15]^+, \ 182 \ (12), \ 169 \ (5), \ 150 \ (100), \ 149$ (21), 133 (12), 122 (44), 111 (6), 105 (50), 77 (21), 43 (87) ppm. C₁₂H₁₆O₄ (224.25): calcd. C 64.27, H 7.19; found C 64.46 H, 7.28.

3-Methyl-5-phenylpent-1-yn-3-yl Acetate (25): A solution of 4-phenyl-2-butanone (30, 0.5 g, 3.38 mmol) in a mixture of dry THF/diethyl ether (1:1, 2 mL) was added dropwise, at room temp., to a solution of ethynylmagnesium bromide (8.5 mL, 4.05 mmol, 0.5 m in THF) in anhydrous diethyl ether (8.5 mL), under argon. The reaction mixture was stirred at room temp. for 12 h. Water (25 mL) was added to quench the reaction, and the mixture was extracted with diethyl ether (3×25 mL). The combined organic layers were washed with brine and dried with anhydrous Na₂SO₄. The solvent was evaporated under vacuum. Purification by flash chromatography (EtOAc/hexane, 3:97) gave 31 (510 mg, 86% yield) which showed spectroscopic data identical to those described in the literature^[39] [¹H NMR (200 MHz, CDCl₃): $\delta = 7.25$ –7.10 (m, 5 H, C₆H₅), 2.82–2.76 (m, 2 H), 2.73 (s, 1 H, OH), 2.43 (s, 1 H), 1.95–1.84 (m, 2 H), 1.47 (s, 3 H, CH₃) ppm].

To a solution of compound **31** (400 mg, 2.3 mmol) in dry pyridine (5 mL), DMAP (56.1 mg, 0.46 mmol) and acetic anhydride (351.9 mg, 0.5 mL, 3.45 mmol) were added. The reaction mixture was stirred overnight at room temp. The solvent was evaporated, and the crude was purified by flash chromatography (EtOAc/hexane, 1:99) to furnish **25** (333 mg, 70% yield) as a colorless oil, which showed spectroscopic data identical to those described in the literature^[51] [1 H NMR (200 MHz, CDCl₃): δ = 7.25–7.10 (m, 5 H, C₆H₅), 2.78–2.69 (m, 2 H), 2.53 (s, 1 H), 2.26–1.98 (m, 2 H), 1.93 (s, 3 H, OCOCH₃), 1.65 (s, 3 H, CH₃) ppm].

4,8-Dimethylnon-7-en-1-yn-4-yl Acetate (26): To a solution of 6methyl-5-heptene-2-one (33, 2.0 g, 15.8 mmol) in a mixture of THF/saturated aqueous solution of NH₄Cl (1:5, 70 mL), zinc dust (6.18 g, 95.2 mmol) was added under argon. The mixture was stirred in an ice bath (0 °C). Thereafter, propargyl bromide (5.33 mL, 47.6 mmol) was added, and the reaction was stirred for 19 h at room temp. The mixture was concentrated, and the crude material was extracted with ethyl acetate (3×10 mL). The organic extract was washed with brine, dried with anhydrous Na2SO4 and concentrated under reduced pressure. Flash column chromatography of the crude material, eluting with EtOAc/hexane (0.5:99.5), gave unreacted 6-methyl-5-heptene-2-one (33, 883 mg) and known compound 34^[56] (211 mg, 12% yield) as a light yellow oil: [¹H NMR (300 MHz, CDCl₃): $\delta = 5.09$ (tm, J = 7.1 Hz, 1 H), 2.41 (d, J = 2.7 Hz, 2 H), 2.16–2.04 (m, 3 H, 2 H, OH), 1.85 (s, 1 H), 1.71 (s, 3 H, CH₃), 1.65 (s, 3 H, CH₃), 1.69–1.58 (m, 2 H), 1.31 (s, 3 H, CH₃) ppm].

To a solution of compound 34 (290 mg, 1.75 mmol) in dry pyridine (2 mL), DMAP (42.7 mg, 0.35 mmol) and acetic anhydride (267.3 mg, 2 mL, 2.62 mmol) were added. The reaction mixture was stirred at room temp. for 16 h. Purification by flash column chromatography (EtOAc/hexane, 0.5:99.5) gave **26** (161.2 mg, 44% yield) as a colorless oil. IR (neat): $\tilde{v} = 3310, 2966, 2925, 2857, 2121,$ 1736, 1450, 1368, 1247 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 5.10 (tm, J = 7.1 Hz, 1 H), 2.75 (dd, J = 12.0, 2.0 Hz, 1 H), 2.55(dd, $J = 12.0, 2.0 \text{ Hz}, 1 \text{ H}), 1.97 \text{ (s, 4 H, OCOCH}_3, H), 2.00–1.85$ (m, 3 H), 1.80–1.65 (m, 1 H), 1.61 (s, 3 H, CH₃), 1.50 (s, 3 H, CH₃), 1.45 (s, 3 H, CH₃) ppm. ¹³C NMR (300 MHz, CDCl₃): $\delta = 170.7$ (OCOCH₃), 132.3 (C), 123.9 (CH), 82.7 (C), 80.5 (C), 70.7 (CH), 38.1 (CH₂), 29.0 (CH₂), 26.0 (CH₃), 23.9 (CH₂), 22.6 (CH₃), 22.5 $(OCOCH_3)$, 17.9 (CH_3) ppm. MS (70 eV): m/z (%) = 208 (9) [M]⁺, 191 (20), 175 (6), 147 (100), 131 (9), 73 (75), 55 (11), 41 (12). $C_{13}H_{20}O_2$ (298.30): calcd. C 74.96, H 9.68; found C 74.83, H 9.72.

Cycloisomerization Products

General Procedure (GP) for the PtCl_n-Catalyzed Cycloisomerization Reaction: To a degassed solution of the precursor in dry toluene (0.025 M), PtCl_n (0.05 equiv.) was added at room temp. under argon. The reaction mixture was stirred at the stated temperature until the reaction was complete. The reaction mixture was filtered, and the solvent was evaporated under vacuum. Purification by flash chromatography gave the corresponding products.

2-Ethenyl-4,4-dimethyl-1-(2-propenyl)-2-cyclopenten-1-ol (2a) and $(1aR^*,3aS^*,4aR^*,5aR^*,5bS^*)$ -octahydro-2,2-dimethyl-3aH-dicyclopropa[a,g]pentalen-3a-ol (3a): Cyclization of precursor 1a (178 mg, 1 mmol) with PtCl₂ (10.4 mg, 0.04 mmol), following the GP (80 °C, 2 h) gave, after purification by flash chromatography (EtOAc/hexane, 5:95), triene 2a (35 mg, 20%) and compound 3a (85 mg, 48%). **2a**: IR (neat): $\tilde{v} = 3400$, 2920, 1630, 1450 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): $\delta = 6.28$ (dd, J = 18.0, 11.4 Hz, 1 H), 5.80 (m, 1 H), 5.69 (s, 1 H), 5.60 (dd, J = 18.0, 2.0 Hz, 1 H trans), 5.18-5.09 (m, 3 H, 2H+H cis), 2.63 (dd, J = 13.7, 8.0 Hz, 1 H), 2.41 (dd, J = 13.7, 8.0 Hz, 1 H), 2.17 (br. s, 1 H), 2.13 (d, J =14.0 Hz, 1 H), 1.74 (d, J = 14.0 Hz, 1 H), 1.35 (s, 3 H, CH₃), 0.97 (s, 3 H, CH₃) ppm. 13 C NMR (100 MHz, CDCl₃): δ = 142.4 (C), 142.2 (CH), 134.2 (CH), 130.4 (CH), 118.6 (CH₂), 116.0 (CH₂), 85.4 (C), 53.5 (CH₂), 44.7 (CH₂), 42.0 (C), 30.7 (CH₃), 28.8 (CH₃) ppm. C₁₂H₁₈O (178.27): calcd. C 80.85, H 10.18; found C 80.62, H 10.23. **3a**: IR (neat): $\tilde{v} = 3370$, 2950, 1460, 1360 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): δ = 2.21 (br. s, 1 H), 2.16 (dd, J = 14.1, 6.6 Hz, 1 H), 1.71 (m, 1 H), 1.67 (dd, J = 14.1, 3.2 Hz, 1 H), 1.60 (d, J = 14.1) 14.2 Hz, 1 H), 1.36 (td, J = 7.9, 4.0 Hz, 1 H), 1.06 (s, 3 H, CH₃), 1.03 (d, J = 14.2 Hz, 1 H), 0.92 (s, 3 H, CH₃), 0.89 (td, J = 7.9,

4.7 Hz, 1 H), 0.85 (dd, J = 8.4, 4.9 Hz, 1 H), 0.78 (dd, J = 8.4, 4.2 Hz, 1 H), 0.69 (dd, J = 4.9, 4.2 Hz, 1 H), 0.50 (dt, J = 4.7, 4.0 Hz, 1 H) ppm. 13 C NMR (125 MHz, CDCl₃): δ = 99.2 (C), 47.2 (CH₂), 45.9 (C), 44.4 (CH₂), 39.3 (C), 35.4 (CH), 30.2 (CH₃), 28.1 (CH₃), 22.2 (CH₂), 20.5 (CH), 20.1 (CH), 11.5 (CH₂) ppm. GCMS (EI): m/z (%) = 178 (6) [M], 145 (100). $C_{12}H_{18}O$ (178.27): calcd. C 80.85, H 10.18; found C 80.65, H 9.98.

(Bromomethyl){[2-ethenyl-4,4-dimethyl-1-(2-propenyl)-2-cyclopenten-1-ylloxy}dimethylsilane (2b) and (Bromomethyl)dimethyl- $\{[(1aR^*, 3aS^*, 4aR^*, 5aR^*, 5bS^*) - \text{octahydro-} 2, 2 - \text{dimethyl-} 3aH - \text{dimet$ cyclopropa[a,g]pentalen-3a-yl]oxy}silane (3b): Cyclization of precursor 1b (308 mg, 0.94 mmol) with $PtCl_2$ (12 mg, 0.047 mmol), following the GP (80 °C, 4 h) gave, after purification by flash chromatography (hexane/CH₂Cl₂, 9:1), triene **2b** (26 mg, 8%) and compound **3b** (188 mg, 61%). **2b**: Oil. IR (neat): $\tilde{v} = 3080$, 2940, 1630, 1440, 1250 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): δ = 6.25 (dd, J = 17.8, 11.2 Hz, 1 H), 5.77 (m, 1 H), 5.61 (s, 1 H), 5.59 (dd, J =18.0, 1.6 Hz, 1 H trans), 5.18-5.02 (m, 3 H, 2 H+H cis), 2.66 (dd, J = 13.4, 7.0 Hz, 1 H), 2.49 (s, 2 H, CH₂-Br), 2.33 (dd, J = 13.4, 7.0 Hz, 1 H), 2.15 (d, J = 13.0 Hz, 1 H), 1.80 (d, J = 13.0 Hz, 1 H), 1.17 (s, 3 H, CH₃), 1.06 (s, 3 H, CH₃), 0.28 [s, 6 H, Si(CH₃)₂] ppm. ¹³C NMR (50 MHz, CDCl₃): δ = 143.0 (C), 140.7 (CH), 134.8 (CH), 130.6 (CH), 117.5 (CH₂), 115.9 (CH₂), 88.5 (C), 52.5 (CH₂), 46.3 (CH₂), 41.7 (C), 30.4 (CH₃), 28.5 (CH₃), 17.6 (CH₂, CH₂-Br), -0.9 [2 CH₃, Si(CH₃)₂] ppm. **3b**: IR (neat): $\tilde{v} = 3060$, 2990, 2960, 2870, 1740, 1450, 1350 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): $\delta = 2.54$ (s, 2 H, CH₂-Br), 2.21 (dd, J = 15.0, 7.4 Hz, 1 H), 1.70 (m, 1 H), 1.69 (d, J = 9.4 Hz, 1 H), 1.63 (d, J = 14.0 Hz, 1 H), 1.37 (td, J = 7.4, 4.0 Hz, 1 H), 1.11 (d, J = 14.0 Hz, 1 H), 1.07 (s, 3 H, CH₃), 0.93 (s, 3 H, CH₃), 0.90 (m, 2 H), 0.76 (m, 2 H), 0.49 (q, J = 4.0 Hz, 1 H), 0.32 [s, 3 H, Si(CH₃)₂], 0.31 [s, 3 H, $Si(CH_3)_2$ ppm. ¹³C NMR (50 MHz, CDCl₃): $\delta = 102.7$ (C), 48.9 (CH₂), 45.9 (C), 45.5 (CH₂), 38.9 (C), 36.0 (CH), 30.3 (CH₃), 28.5 (CH₃), 22.5 (CH₂), 20.9 (2 CH), 18.3 (CH₂, CH₂-Br), 13.0 (CH₂), -0.7 [2 CH₃, Si(CH₃)₂] ppm. GCMS (EI): m/z (%) = 330 (21) $[M(^{81}Br)]$, 328 (21) $[M(^{79}Br)]$, 145 (100). $C_{15}H_{25}BrOSi$ (329.36): calcd. C 54.70, H 7.65; found C 55.00 H, 7.87.

 $(1aR^*,3aS^*,4aR^*,5aR^*,5bS^*)$ -Octahydro-3a-methoxy-2,2-dimethyl-1H-dicyclopropa[a,g]pentalene (3c): Cyclization of precursor 1c (97 mg, 0.5 mmol) with PtCl₂ (6.5 mg, 0.025 mmol), following the GP (80 °C, 3 h) gave, after purification by flash chromatography (Et₂O/pentane: 2:98), compound 3c (71 mg, 73%). Alternatively, cyclization of precursor 1c (137 mg, 0.71 mmol) was conducted with PtCl₄ (12 mg, 0.036 mmol), following the same procedure (80 °C, 1 h) and gave, after purification by flash chromatography (Et₂O/pentane, 2:98), compound **3c** (105 mg, 77%). **3c**: IR (neat): $\tilde{v} = 2950, 1480, 1380 \text{ cm}^{-1}. {}^{1}\text{H NMR } (200 \text{ MHz}, \text{CDCl}_{3}): \delta = 3.28$ (s, 3 H, OMe), 2.36 (dd, J = 14.3, 7.0 Hz, 1 H), 1.61 (m, 1 H), 1.52 (dd, J = 14.3, 2.5 Hz, 1 H), 1.52 (d, J = 14.3 Hz, 1 H), 1.33 (td, J)= 7.4, 3.4 Hz, 1 H), 1.08 (d, J = 14.3 Hz, 1 H), 1.06 (s, 3 H, CH₃), 0.90 (s, 3 H, CH₃), 0.89 (td, J = 8.4, 4.4 Hz, 1 H), 0.82-0.70 (m, 3 H), 0.47 (q, J = 4.0 Hz, 1 H) ppm. ¹³C NMR (50 MHz, CDCl₃): δ = 104.0 (C), 51.9 (CH₃, OMe), 46.2 (CH₂), 43.8 (C), 39.2 (C, CH₂), 35.8 (CH), 30.7 (CH₃), 28.3 (CH₃), 22.1 (CH₂), 21.7 (CH), 21.2 (CH), 12.2 (CH₂) ppm. C₁₃H₂₀O (192.30): C 81.20, H 10.48; found C 81.07, H 10.59.

(1aR*,3aS*,4aR*,5aR*,5bR*)-Octahydro-3a-methoxy-2,2,4a-tri-methyl-1*H*-dicyclopropa[*a*,*g*]pentalene (3d): Cyclization of precursor 1d (86 mg, 0.4 mmol) with PtCl₂ (5 mg, 0.02 mmol), following the GP (80 °C, 3 h) gave, after purification by flash chromatography (Et₂O/pentane, 2:98), compound 3d (29 mg, 34%). Alternatively, cyclization of precursor 1d (80 mg, 0.39 mmol) was conducted with

PtCl₄ (7 mg, 0.02 mmol), following the same procedure (80 °C, 2 h) and gave, after purification by flash chromatography (Et₂O/pentane, 2:98), compound **3d** (32 mg, 40%). **3d**: IR (neat): \tilde{v} = 3040, 2950, 1460 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): δ = 3.28 (s, 3 H, OMe), 2.19 (d, J = 14.6 Hz, 1 H), 1.72 (d, J = 14.6 Hz, 1 H), 1.53 (d, J = 13.8 Hz, 1 H), 1.30 (s, 3 H, CH₃), 1.13 (d, J = 13.8 Hz, 1 H), 1.12 (m, 1 H), 1.09 (s, 3 H, CH₃), 0.94 (s, 3 H, CH₃), 0.91 (m, 1 H), 0.85–0.70 (m, 3 H), 0.64 (t, J = 3.7 Hz, 1 H) ppm. ¹³C NMR (50 MHz, CDCl₃): δ = 103.2 (C), 52.3 (CH₃, OMe), 46.5 (CH₂), 45.3 (C), 45.2 (CH₂), 39.6 (C), 36.0 (CH), 31.0 (CH₃), 30.0 (C), 28.9 (CH, CH₂), 28.6 (CH₃), 23.8 (CH₃), 12.8 (CH₂) ppm.

 $(1aR^*,3aS^*,4aR^*,5aR^*,5bS^*)$ -Octahydro-3aH-dicyclopropa[a,g]pentalen-3a-ol (3e): Cyclization of precursor 1e (83 mg, 0.5 mmol) with PtCl₂ (6.5 mg, 0.025 mmol), following the GP (80 °C, 2 h) gave, after purification by flash chromatography (Et₂O/pentane, 2:98), compound 3e (45 mg, 53%). Alternatively, cyclization of precursor 1e (102 mg, 0.62 mmol) was conducted with PtCl₄ (10 mg, 0.03 mmol), following the same procedure (80 °C, 2 h) and gave, after purification by flash chromatography (Et₂O/pentane, 5:95), compound **3e** (20 mg, 20%). IR (neat): $\tilde{v} = 3060$, 2930, 1460 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 3.33$ (s, 3 H, OMe), 2.30 (dd, J = 14.6, 7.5 Hz, 1 H), 1.80 (m, 1 H), 1.70–1.60 (m, 3 H), 1.38 (td, J = 7.6, 3.3 Hz, 1 H), 1.29 (dd, J = 14.6, 3.0 Hz, 1 H), 1.21 (m, 1 H), 1.02 (m, 1 H), 0.93–0.83 (m, 3 H), 0.39 (m, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 104.5$ (C), 52.2 (CH₃, OMe), 41.3 (C), 37.7 (CH₂), 30.0 (CH₂), 26.2 (CH₂), 22.98 (CH₂), 22.97 (CH), 20.9 (CH), 20.3 (CH), 12.7 (CH₂) ppm.

p-Nitrobenzoate 4: To a solution of compound 3a (89 mg, 0.5 mmol, 1 equiv.) in CH₂Cl₂ (2.5 mL) was added 4-nitrobenzoyl chloride (371 mg, 2 mmol, 4 equiv.). The mixture was heated to 50 °C overnight and quenched with 10% aqueous HCl. The aqueous layer was extracted with Et₂O. The combined organic layers were washed with brine and dried with MgSO₄. Purification by flash chromatography on silica gel (pentane/CH₂Cl₂, 9:1) gave the pure ester 4 (51 mg, 31% yield). The solid product was recrystallized from hexane/Et₂O. 4: Solid, m.p. 135 °C. IR (in CDCl₃ solution): $\tilde{v} = 3040$, 2940, 1700, 1600, 1520, 1450 cm⁻¹. ¹H NMR (200 MHz, CDCl₃): $\delta = 8.23$ (m, 4 H, Ar), 2.74 (dd, J = 15.7, 7.4 Hz, 1 H), 2.26 (d, J = 14.8 Hz, 1 H), 1.83 (dd, J = 15.7, 2.5 Hz, 1 H), 1.64 (m, 1 H), 1.49 (td, J = 7.4, 3.4 Hz, 1 H), 1.16 (d, J =14.8 Hz, 1 H), 1.14 (s, 3 H, CH₃), 1.06 (m, 1 H), 0.93 (s, 3 H, CH₃), 0.96-0.84 (m, 3 H), 0.58 (q, J = 3.9 Hz, 1 H) ppm. ¹³C NMR (50 MHz, CDCl₃): δ = 165.0 (C, C=O), 150.7 (C, Ar), 137.5 (C, Ar), 131.0 (2 CH, Ar), 123.8 (2 CH, Ar), 109.9 (C), 46.3 (CH₂), 45.2 (C), 41.9 (CH₂), 40.2 (C), 35.2 (CH), 30.7 (CH₃), 28.9 (CH₃), 22.7 (CH₂), 21.0 (2 CH), 12.9 (CH₂) ppm.

 $(1aR^*,3aS^*,4aR^*,5aR^*,5bR^*)$ -Octahydro-3a-allyloxy-2,2,4a-trimethyl-1*H*-dicyclopropa[*a*,*g*]pentalene (3f) and Tetracycle 5: Cyclization of 1f with PtCl₂ (85 mg, 0.39 mmol), following the GP (16 h at room temp.) gave, after purification by flash chromatography (CH₂Cl₂/hexane, 1:4), compounds **3f** (38 mg, 44% yield) and **5** (31 mg, 36% yield). **3f**: Oil. IR (neat): $\tilde{v} = 3060$, 2950, 1650, 1450, 1360, 1290 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): δ = 5.97 (dddd, J = 17.3, 10.5, 5.6, 5.4 Hz, 1 H), 5.30 (dq, J = 17.3, 1.7 Hz, 1 H trans), 5.12 (dq, J = 10.5, 1.3 Hz, 1 H cis), 4.05 (m, 2 H, 2H), 2.37 (dd, J = 15.1, 7.4 Hz, 1 H), 1.64 (m, 3 H), 1.38 (td, J = 7.5, 3.6 Hz,1 H), 1.19 (d, J = 13.9 Hz, 1 H), 1.10 (s, 3 H, CH₃), 0.95 (m, 1 H), 0.94 (s, 3 H, CH₃), 0.86 (m, 2 H), 0.78 (m, 1 H), 0.50 (q, J = 3.8 Hz, 1 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 136.7 (CH), 116.0 (CH₂), 104.5 (C), 65.9 (CH₂), 47.6 (CH₂), 44.2 (C), 40.9 (CH₂), 39.5 (C), 36.5 (CH), 31.1 (CH₃), 28.6 (CH₃), 22.3 (CH, CH₂), 21.6 (CH), 12.8 (CH₂) ppm. MS (70 eV): m/z (%) = 218 (1) [M]⁺, 207

(6), 183 (16), 168 (51), 105 (21), 91 (100), 79 (12), 77 (29). C₁₅H₂₂O (218.33): calcd. C 82.52, H 10.16; found C 82.67, H 10.38. 5: IR (neat): $\tilde{v} = 3060$, 2950, 1650, 1450, 1360, 1290 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 5.95 (m, 1 H), 5.13 (d, J = 10.1 Hz, 1 H cis), 5.12 (d, J = 17.2 Hz, 1 H trans), 4.26 (dd, J = 11.8, 7.3 Hz, 1 H), 3.55 (dd, J = 11.8, 7.0 Hz, 1 H), 3.13 (m, 1 H), 2.37 (dd, J =14.4, 9.2 Hz, 1 H), 1.62 (d, J = 13.5 Hz, 1 H), 1.50 (dd, J = 8.0, 3.2 Hz, 1 H), 1.45 (m, 1 H), 1.34 (m, 1 H), 1.19 (s, 3 H, CH₃), 0.98 (s, 3 H, CH₃), 0.90 (d, J = 13.5 Hz, 1 H), 0.88 (m, 1 H), 0.49 (t, J= 4.1 Hz, 1 H, 0.39 (q, J = 4.6 Hz, 1 H), 0.21 (dd, J = 7.9, 4.5 Hz,1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 136.1$ (CH), 117.0 (CH₂), 81.0 (C), 66.4 (CH₂), 43.4 (CH₂), 42.3 (CH₂), 38.5 (C), 35.5 (CH), 34.7 (C), 29.0 (CH₃), 28.5 (CH₃), 16.4 (CH₂), 15.0 (CH₂), 13.5 (CH), 10.6 (CH) ppm. MS (70 eV): m/z (%) = 218 (1) [M]⁺, 183 (5), 161 (9), 135 (11), 105 (22), 97 (44), 69 (87), 43 (88). C₁₅H₂₂O (218.33): calcd. C 82.52, H 10.16; found C 82.47, H 10.27.

Ester 6: A solution of compound 5 (82 mg, 0.4 mmol, 1 equiv.) in CH₂Cl₂/methanol (1:1, 8 mL) was cooled to -78 °C. Ozone was bubbled into the solution until a persistent blue color was perceived. The excess of ozone was then removed by bubbling argon into the reaction vessel. Sodium borohydride (29 mg, 0.8 mmol, 2 equiv.) was added, and the mixture was slowly warmed to 0 °C. After 1 h of stirring at 0 °C, the reaction was quenched with a saturated ammonium chloride solution for 30 min. The aqueous layer was extracted with Et₂O. The combined organic layers were washed with brine, dried with MgSO₄, and the solvents were removed under vacuum. To a cold (0 °C) solution of the resulting crude alcohol (0.4 mmol, 1 equiv.) in CH₂Cl₂ (3 mL) were added triethylamine (110 μL, 0.8 mmol, 2 equiv.), DMAP (10 mg, 0.08 mmol, 0.2 equiv.) and 4-nitrobenzoyl chloride (82 mg, 0.44 mmol, 1.1 equiv.). After a 15 min of stirring at room temp., the reaction was quenched with a saturated ammonium chloride solution. The aqueous layer was extracted with Et₂O. The combined organic layers were washed with brine and dried with MgSO₄. Purification by flash chromatography on silica gel (Et₂O/pentane, 2:8) gave the desired ester 6 (87 mg, 65% yield over 2 steps). The solid product was recrystallized from hexane/Et₂O. Solid, m.p. 94 °C. IR (film): $\tilde{v} = 2960, 1720, 1600, 1520, 1280 \text{ cm}^{-1}$. ¹H NMR (400 MHz, CDCl₃): $\delta = 8.30$ (d, J = 9.1 Hz, 2 H, Ar), 8.23 (d, J = 9.1 Hz, 2 H, Ar), 4.58 (m, 2 H), 4.25 (dd, J = 11.9, 7.3 Hz, 1 H), 3.51 (dd, J= 11.9, 7.6 Hz, 1 H), 2.78 (m, 1 H), 2.17 (m, 1 H), 1.60 (d, J = 13.6 Hz, 1 H), 1.53 (dd, J = 8.1, 3.3 Hz, 1 H), 1.47 (m, 1 H), 1.37 (m, 1 H), 1.22 (s, 3 H, CH₃), 1.03 (d, J = 13.6 Hz, 1 H), 1.00 (s, 3 H, CH₃), 0.89 (td, J = 7.6, 5.1 Hz, 1 H), 0.50 (t, J = 4.0 Hz, 1 H), 0.40 (q, J = 4.6 Hz, 1 H), 0.23 (dd, J = 8.1, 4.8 Hz, 1 H) ppm.NMR (100 MHz, CDCl₃): δ = 164.8 (C, C=O), 150.4 (C, Ar), 136.0 (C, Ar), 130.7 (2 CH, Ar), 123.5 (2 CH, Ar), 79.7 (C), 66.7 (CH₂), 63.7 (CH₂), 42.9 (CH₂), 38.5 (C), 37.3 (CH₂), 35.3 (CH), 34.5 (C), 29.0 (CH₃), 28.2 (CH₃), 16.5 (CH₂), 14.7 (CH₂), 13.4 (CH), 10.5 (CH) ppm.

5-(Hex-5-enyl)-5-methoxy-3,3-dimethyl-1-vinylcyclopentene (8) and 1-(Hex-5-enyl)-1-methoxy-4,4-dimethyl-2,3-dimethylenecyclopentane (9): Cyclization of precursor **7a** (92 mg, 0.4 mmol) with PtCl₂ (5 mg, 0.02 mmol), following the GP (50 °C, 6 h) gave, after purification by flash chromatography on silica gel (pentane/ CH₂Cl₂, 9:1), cyclopentene **8** (55 mg, 60%) and cyclopentane **9** (15 mg, 16%). **8**: IR (neat): \tilde{v} = 3080, 2930, 1640, 1465 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 6.29 (dd, J = 17.8, 11.2 Hz, 1 H), 5.82 (ddt, J = 17.1, 10.5, 6.6 Hz, 1 H), 5.70 (s, 1 H), 5.59 (dd, J = 17.8, 1.8 Hz, 1 H), 5.09 (dd, J = 11.2, 1.8 Hz, 1 H), 5.01 (dd, J = 17.1, 1.5 Hz, 1 H), 4.94 (dd, J = 10.5, 1.5 Hz, 1 H), 3.13 (s, 3 H, O–CH₃), 2.09 (m, 2 H, CH₂), 1.91 (d, J = 14.8 Hz, 1 H), 1.85 (d, J = 14.8 Hz, 1 H), 1.42–1.25 (m, 6 H, 3CH₂), 1.15 (s, 3 H, CH₃),

1.10 (s, 3 H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 144.3$ (CH), 139.9 (C), 139.4 (CH), 131.3 (CH), 115.7 (CH₂), 114.6 (CH₂), 92.7 (C), 50.6 (CH₃, O–Me), 46.6 (CH₂), 41.7 (C), 39.8 (CH₂), 34.1 (CH₂), 30.1 (CH₃), 29.8 (CH₂), 29.7 (CH₃), 24.1 (CH₂) ppm. C₁₆H₂₆O (234.38): calcd. C 81.99, H 11.18; found C 81.85, H 11.38. **8D**, deuterated at C-2: ¹H NMR (400 MHz, CDCl₃): the same spectrum as that of 8, except no signal at $\delta = 6.29$ ppm and $\delta = 5.59$ (d, J = 1.8 Hz, 1 H), 5.09 (s, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): same spectrum as that of 8, except $\delta = 131.3$ (t, J = 22 Hz) ppm; deuterated at C-1: ¹H NMR (400 MHz, CDCl₃): the same spectrum as that of 8, except no signal at δ = 5.59 ppm and $\delta = 6.29$ (d, J = 11.4 Hz, 1 H), 5.09 (d, J = 11.4 Hz, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): the same spectrum as that of **8**, except $\delta = 115.7$ (t, J = 24 Hz) ppm. ESI-MS: m/z (%) = 274 (30) [M + K]⁺, 258 (100) [M + Na]⁺, 236 (10) [M + H]⁺. 9: IR (neat): $\tilde{v} = 3080$, 2930, 1640, 1460 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 5.82$ (ddt, J = 17.1, 10.5, 6.6 Hz, 1 H), 5.58 (s, 1 H), 5.40 (s, 1 H), 5.01 (dd, J = 17.1, 1.5 Hz, 1 H), 4.94 (dd, J = 10.5, 1.5 Hz, 1 H), 4.93 (s, 1 H), 4.85 (s, 1 H), 3.07 (s, 3 H, O-CH₃), 2.09 (m, 2 H), 1.91 (d, J = 13.6 Hz, 1 H), 1.91 (m, 1 H), 1.59 (d, J= $13.6 \,\mathrm{Hz}$, 1 H), $1.46-1.25 \,\mathrm{(m, 5 H)}$, $1.23 \,\mathrm{(s, 3 H, CH_3)}$, $1.13 \,\mathrm{(s, 3 H, CH_3)}$ H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 159.2 (C), 150.4 (C), 139.3 (CH), 114.7 (CH₂), 107.5 (CH₂), 104.3 (CH₂), 84.4 (C), 50.7 (CH₃), 49.7 (CH₂), 40.1 (C), 34.7 (CH₂), 34.2 (CH₂), 31.7 (CH₃), 31.1 (CH₃), 29.8 (CH₂), 23.7 (CH₂) ppm. **9D**: ¹H NMR (400 MHz, CDCl₃): the same spectrum as that of 9, except no signal at δ = 4.85 ppm. ¹³C NMR (100 MHz, CDCl₃): the same spectrum as that of **9**, except $\delta = 104.3$ (t, J = 23 Hz) ppm.

4-(2,2-Dimethylhex-5-enyl)-4-methoxy-3-methylenecyclohexene (10) and 1-(2,2-Dimethylhex-5-enyl)-1-methoxy-2,3-dimethylenecyclopentane (11): Cyclization of precursor 7b (80 mg, 0.34 mmol) with PtCl₂ (5 mg, 0.02 mmol), following the GP (50 °C, 6 h) gave, after purification by flash chromatography on silica gel (pentane/Et₂O, 98:2), cyclohexene 10 (36 mg, 45%) and cyclopentane 11 (11 mg, 14%). 10: IR (neat): $\tilde{v} = 3080$, 2940, 1640, 1460 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 6.08$ (d, J = 10.2 Hz, 1 H), 5.79 (ddt, J =16.8, 10.2, 6.6 Hz, 1 H), 5.67 (m, 1 H), 5.13 (s, 1 H), 4.97 (d, J =16.8 Hz, 1 H), 4.95 (s, 1 H), 4.88 (d, J = 10.2 Hz, 1 H), 3.15 (s, 3 H), 2.25 (m, 2 H), 2.03–1.90 (m, 3 H), 1.76 (dt, J = 12.2, 4.1 Hz, 1 H), 1.55 (m_{AB}, 2 H), 1.42 (m, 1 H), 1.35 (m, 1 H), 1.00 (s, 3 H, CH₃), 0.97 (s, 3 H, CH₃) ppm. 13 C NMR (100 MHz, CDCl₃): δ = 146.1 (C), 140.5 (CH), 130.2 (CH), 128.2 (CH), 113.8 (CH₂), 112.1 (CH₂), 79.2 (C), 49.4 (CH₃), 46.0 (CH₂), 44.0 (CH₂), 34.5 (C), 30.9 (CH₂), 29.2 (2 CH₃), 29.0 (CH₂), 25.3 (CH₂) ppm. 11: IR (neat): \tilde{v} = 3080, 2930, 1640, 1450 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 5.83 (ddt, J = 16.8, 10.2, 6.6 Hz, 1 H), 5.59 (s, 1 H), 5.41 (t, J =2.5 Hz, 1 H), 5.02 (d, J = 10.2 Hz, 1 H), 4.97 (d, J = 16.8 Hz, 1 H), 4.95 (s, 1 H), 4.92 (t, J = 2.0 Hz, 1 H), 3.05 (s, 3 H), 2.55 (m, 1 H), 2.36 (m, 1 H), 2.14 (ddd, J = 12.7, 8.1, 3.0 Hz, 1 H), 2.05(m, 2 H), 1.98 (d, J = 15.2 Hz, 1 H), 1.72 (m, 1 H), 1.43 (m, 2 H),1.29 (d, J = 15.2 Hz, 1 H), 1.05 (s, 3 H, CH₃), 1.04 (s, 3 H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 150.7 (C), 147.5 (C), 139.8 (CH), 113.8 (CH₂), 106.6 (CH₂), 105.0 (CH₂), 86.0 (C), 50.0 (CH₃), 43.8 (CH₂), 42.8 (CH₂), 36.2 (CH₂), 34.0 (C), 29.8 (CH₂), 28.8 (CH₂), 28.6 (CH₃), 28.1 (CH₃) ppm.

5-Methoxy-3,3,5-trimethyl-1-vinylcyclopentene (12) and 1-Methoxy-1,4,4-trimethyl-2,3-dimethylenecyclopentane (13): Cyclization of precursor 7c (166 mg, 1.0 mmol) with PtCl₂ (14 mg, 0.05 mmol) and styrene (580 μL, 5 mmol), following the GP (80 °C, 1 h) gave, after purification by flash chromatography on silica gel (pentane/ CH₂Cl₂, 9:1), a mixture of 12 and 13 (3:1, 82 mg, 49% yield). 12: ¹H NMR (400 MHz, CDCl₃): δ = 6.25 (dd, J = 18.2, 11.8 Hz, 1 H), 5.62 (s, 1 H), 5.60 (d, J = 18.2 Hz, 1 H), 5.09 (d, J = 11.8 Hz,

1 H), 3.11 (s, 3 H, OMe), 2.06 (d, J = 14.1 Hz, 1 H), 1.67 (d, J = 14.1 Hz, 1 H), 1.47 (s, 3 H, CH₃), 1.12 (s, 3 H, CH₃), 1.08 (s, 3 H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 143.6 (CH), 141.1 (C), 131.1 (CH), 115.8 (CH₂, CH=*C*H₂), 89.3 (C), 50.7 (CH₃, OMe), 49.7 (CH₂), 41.6 (C), 29.8 (CH₃), 29.7 (CH₃), 28.3 (CH₃) ppm. 13: ¹H NMR (400 MHz, CDCl₃): δ = 5.54 (s, 1 H), 5.39 (s, 1 H), 4.98 (s, 1 H), 4.84 (1s, 1 H), 3.11 (s, 3 H, OMe), 1.97 (d, J = 13.6 Hz, 1 H), 1.58 (d, J = 13.6 Hz, 1 H), 1.47 (s, 3 H, CH₃), 1.33 (s, 3 H, CH₃), 1.22 (s, 3 H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 158.7 (C), 107.2 (CH₂), 104.3 (CH₂), 82.2 (C), 51.9 (CH₂), 50.9 (CH₃, OMe), 40.2 (C), 31.1 (CH₃), 31.0 (CH₃), 22.9 (CH₃) ppm.

4-Butyl-4-methoxy-3-methylenecyclohexene (14) and 1-Butyl-1methoxy-2,3-dimethylenecyclopentane (15): Cyclization of precursor 7d (180 mg, 1.0 mmol, 1 equiv.) following the GP (40 °C, 30 h) gave, after purification by flash chromatography on silica gel (pentane/ CH₂Cl₂, 95:5), a mixture of **14** and **15** (**14/15**, 4:1, 38 mg, 21% yield). **14**: ¹H NMR (400 MHz, CDCl₃): $\delta = 6.07$ (d, J = 9.6 Hz, 1 H), 5.71 (m, 1 H), 5.06 (s, 1 H), 4.98 (s, 1 H), 3.17 (s, 3 H, OMe), 2.56–2.51 (m, 1 H), 2.35–2.23 (m, 1 H), 1.98–1.93 (m, 1 H), 1.62– 1.56 (m, 1 H), 1.38–1.24 (br. m, 6 H), 0.86–0.93 (m, 3 H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 144.1$ (C), 129.3 (CH), 128.4 (CH), 111.5 (CH₂), 77.2 (C), 49.4 (CH₃, OMe), 36.1 (CH₂), 31.9 (CH₂), 29.7 (CH₂), 25.0 (CH₂), 24.1 (CH₂), 14.1 (CH₃) ppm. **15**: ¹H NMR (400 MHz, CDCl₃): 5.59 (s, 1 H), 5.41 (s, 1 H), 4.93 (s, 1 H), 4.91 (s, 1 H), 3.08 (s, 3 H, OMe), 2.56–2.51 (m, 1 H), 2.29 (m, 1 H), 1.95 (m, 1 H), 1.59 (m, 1 H), 1.38-1.24 (m, 6 H), 0.89 (m, 3 H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): δ = 148.8 (C), 147.8 (C), 106.2 (CH₂), 105.0 (CH), 85.5 (C), 49.6 (CH₃, OMe), 30.3 (CH₂), 29.7 (CH₂), 29.4 (CH₂), 25.8 (CH₂), 23.2 (CH₂), 14.2 (CH_3) ppm.

Methyl 2-(5-Allyl-5-methoxy-3,3-dimethylcyclopent-1-enyl)acrylate (16) and Methyl (2-Allyl-2-methoxy-4,4-dimethyl-5-methylenecyclopentylidene)acetate (17): Cyclization of precursor 7f (125 mg, 0.5 mmol, 1 equiv.) following the GP (80 °C, 1 h) gave, after purification by flash chromatography on silica gel (pentane/CH₂Cl₂, 9:1), a mixture of 16 and 17 (16/17, 2.5:1, 80 mg, 64% yield). 16: IR (neat): $\tilde{v} = 3076$, 2954, 2933, 2864, 1726, 1639 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 6.21$ (s, 1 H), 6.20 (s, 1 H), 6.07 (s, 1 H), 5.67 (m, 1 H), 5.02 (m, 2 H), 3.77 (s, 3 H, CO₂Me), 3.12 (s, 3 H, OMe), 2.50 (m, 2 H), 1.88 (m, 2 H), 1.13 (s, 3 H, CH₃), 1.09 (s, 3 H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 167.7$ (C=O), 148.4 (CH), 134.6 (C), 134.4 (CH), 133.0 (C), 124.2 (CH₂, $CH = CH_2$), 117.8 (CH_2 , $CH = CH_2$), 93.5 (C), 52.0 (CH_3), 50.3 (CH₃), 44.9 (CH₂), 43.8 (CH₂), 41.5 (C), 29.4 (CH₃), 28.9 (CH₃) ppm. 17: IR (neat): $\tilde{v} = 3076$, 2954, 2933, 2864, 1726, 1639 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 6.56 (s, 1 H), 5.85 (m, 1 H), 5.72 (s, 1 H), 5.34 (s, 1 H), 5.12 (m, 2 H), 3.77 (s, 3 H, CO₂Me), 3.04 (s, 3 H, OMe), 2.74 (m, 1 H), 2.26 (m, 1 H), 1.88 (m, 2 H), 1.19 (s, 3 H, CH₃), 1.05 (s, 3 H, CH₃) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 167.7 \text{ (C=O)}, 149.2 \text{ (CH)}, 139.1 \text{ (C)}, 134.0 \text{ (CH)}, 126.3 \text{ (C)},$ 117.7 (CH₂, CH=CH₂), 115.1 (CH₂, CH=CH₂), 76.5 (C), 51.7 (CH₃), 49.7 (CH₃), 44.7 (CH₂), 38.3 (CH₂), 33.1 (C), 31.2 (CH₃), 29.9 (CH₃) ppm.

5,5-Dimethyl-3-(2-propenyl)bicyclo[4.1.0]hept-2-en-2-yl 4-Nitroben-zoate (18a): Cyclization of **1g** (228 mg, 0.7 mmol) with PtCl₂ (9 mg, 0.035 mmol) following the GP (2 h at 80 °C) gave, after purification by flash chromatography (EtOAc/hexane, 2:98), a mixture of **18a** and **19a** (95:5, 183 mg, 80% yield). **18a**: solid, m.p. 73–75 °C. IR (KBr): $\tilde{v} = 3115, 3000, 2950, 2850, 1725, 1610, 1530 \text{ cm}^{-1}$. ¹H NMR (200 MHz, CDCl₃): $\delta = 8.33$ (m, 4 H, Ar), 5.63 (m, 1 H), 4.99 (dd, J = 17.2, 1.6 Hz, 1 H *trans*), 4.98 (dd, J = 11.0, 1.6 Hz, 1 H *cis*),

2.72 (dd, J = 14.6, 6.0 Hz, 1 H), 2.60 (dd, J = 14.6, 6.0 Hz, 1 H), 1.82 (d, J = 16.4 Hz, 1 H), 1.68 (d, J = 16.4 Hz, 1 H), 1.37 (td, J = 8.4, 4.2 Hz, 1 H), 1.15 (s, 3 H, CH₃), 1.09 (s, 3 H, CH₃), 1.06–0.76 (m, 3 H) ppm. ¹³C NMR (50 MHz, CDCl₃): $\delta = 163.3$ (C, C=O), 150.8 (C, Ar), 143.6 (C), 135.4 (C, Ar), 135.0 (CH), 131.1 (2 CH, Ar), 123.8 (2 CH, Ar), 116.2 (CH₂), 115.6 (C), 38.2 (CH₂), 35.0 (CH₂), 29.6 (CH₃), 28.5 (CH₃), 28.1 (C), 26.9 (CH), 13.7 (CH), 10.2 (CH₂) ppm. GCMS (EI): m/z (%) = 327 (4) [M], 150 (100). C₁₉H₂₁NO₄ (327.37): calcd. C 69.71, H 6.47, N 4.28; found C 70.02, H 6.65, N 4.48. **19a**: characteristic signals: ¹H NMR (200 MHz, CDCl₃): $\delta = 5.79$ (dd, J = 17.4, 10.6 Hz, 1 H), 0.99 (s, 6 H, 2 CH₃), 0.39 (m, 1 H) ppm.

3-Allyl-5,5-dimethylbicyclo[4.1.0]hept-2-en-2-yl Acetate (18b): Cyclization of 1h (230 mg, 1.04 mmol) with PtCl₂ (14 mg, 0.05 mmol) following the GP (2 h at 80 °C) gave, after purification by flash chromatography (Et₂O/PE, 5:95) a mixture of 18b and 19b (18b/ **19b**, 97:3, 209 mg, 91% yield). **18b**: IR (neat): $\tilde{v} = 3090$, 2950, 1750, 1690, 1640, 1450, 1360 cm⁻¹. 1 H NMR (200 MHz, CDCl₃): δ = 5.56 (ddt, J = 17.0, 10.0, 7.0 Hz, 1 H), 4.95 (d, J = 17.0 Hz, 1 H trans), 4.93 (d, J = 10.0 Hz, 1 H cis), 2.63 (dd, J = 14.8, 6.4 Hz, 1 H), 2.50 (dd, J = 14.8, 6.4 Hz, 1 H), 2.13 (s, 3 H, OAc), 1.61 (m_{AB}, 2 H), 1.21 (m, 1 H), 1.05 (s, 3 H, CH₃), 0.97 (s, 3 H, CH₃), 0.96 (m, 1 H), 0.77 (m, 1 H), 0.66 (m, 1 H) ppm. ¹³C NMR (50 MHz, CDCl₃): $\delta = 169.4$ (C, C=O), 143.4 (C), 135.4 (CH), 115.8 (CH₂), 114.6 (C), 37.9 (CH₂), 34.9 (CH₂), 29.5 (CH₃), 28.5 (CH₃), 28.0 (C), 26.6 (CH), 20.8 (CH₃, OAc), 13.6 (CH), 9.8 (CH₂) ppm. C₁₄H₂₀O₂ (220.31): C 76.33, H 9.15; found C 76.29, H 9.28. 19b: characteristic signals: ¹H NMR (200 MHz, CDCl₃): $\delta = 5.76$ (dd, J = 17.2, 10.5 Hz, 1 H), 2.12 (s, 3 H, OAc), 0.94 (s, 3 H, CH₃), 0.92 (s, 3 H, CH₃), 0.21 (m, 1 H) ppm.

3-Methallyl-5,5-dimethylbicyclo[4.1.0]hept-2-en-2-yl 4'-Nitrobenzoate (18c): Cyclization of 1i (243 mg, 0.7 mmol) with PtCl₂ (9 mg, 0.034 mmol) following the GP (2 h at 80 °C) gave, after purification by flash chromatography (EtOAc/hexane, 1:99), a mixture of 18c and 19c (18c/19c, 7:3, 213 mg, 88% yield). 18c: Solid, m.p. 57-59 °C. IR (KBr): $\tilde{v} = 3110, 3070, 2960, 1730, 1610, 1525, 1350$ cm⁻¹. ¹H NMR (200 MHz, CDCl₃): δ = 8.31 (m, 4 H, Ar), 4.72, (s, 1 H), 4.67 (s, 1 H), 2.63 (s, 2 H), 1.71 (m_{AB}, 2 H), 1.63 (s, 3 H, CH_3), 1.39 (td, J = 8.4, 4.4 Hz, 1 H), 1.13 (s, 3 H, CH_3), 1.09 (s, 3 H, CH₃), 1.12 (m, 1 H), 0.94–0.75 (m, 2 H) ppm. ¹³C NMR (50 MHz, CDCl₃): $\delta = 163.2$ (C, C=O), 150.8 (C, Ar), 144.2 (C), 142.5 (C), 135.4 (C, Ar), 131.1 (2 CH, Ar), 123.8 (2 CH, Ar), 115.3 (C), 112.3 (CH₂), 39.1 (CH₂), 38.0 (CH₂), 29.6 (CH₃), 28.7 (CH₃), 28.0 (C), 27.0 (CH), 22.0 (CH₃), 13.6 (CH), 10.2 (CH₂) ppm. GCMS (EI): m/z (%) = 341 (10) [M], 150 (100). $C_{20}H_{23}NO_4$ (321.40): calcd. C 70.36, H 6.79, N 4.10; found C 70.65, H 7.00, N 4.40. **19c**: characteristic signals: ¹H NMR (200 MHz, CDCl₃): δ = 5.78 (dd, J = 17.4, 10.6 Hz, 1 H), 4.90 (d, J = 10.6 Hz, 1 H), 4.86 $(d, J = 17.4 \text{ Hz}, 1 \text{ H}), 1.30 \text{ (s, 3 H, CH}_3), 0.98 \text{ (s, 3 H, CH}_3), 0.96$ (s, 3 H, CH₃), 0.51 (t, J = 3.5 Hz, 1 H) ppm.

3-Allylbicyclo[4.1.0]hept-2-en-2-yl Acetate (18d): Cyclization of **1j** (192 mg, 1.0 mmol) with PtCl₂ (14 mg, 0.05 mmol) following the GP (1 h at 80 °C) gave, after purification by simple filtration through silica gel, compound **18d** (178 mg, 93% yield). Alternatively, cyclization of precursor **1j** (192 mg, 1.0 mmol) was conducted with PtCl₄ (17 mg, 0.05 mmol), following the same procedure (80 °C, 2 h) and gave compound **18d** (190 mg, 98%) as an oil. IR (neat): $\tilde{v} = 3076$, 3006, 2977, 2922, 2859, 1754, 1638 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): $\delta = 5.62$ (m, 1 H), 4.91 (m, 2 H), 2.62 (m, 2 H), 2.17 (s, 3 H), 1.96–1.67 (m, 4 H), 1.42 (m, 1 H), 1.20 (m, 1 H), 0.75 (m, 2 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 169.7$ (C, C=O), 144.4 (C), 135.7 (CH), 116.8 (C), 115.9 (CH₂),

35.1 (CH₂), 23.8 (CH₂), 21.2 (CH₃, OAc), 19.7 (CH₂), 14.8 (CH), 12.5 (CH), 10.2 (CH₂) ppm.

3-Allyl-5,5-dimethylbicyclo[4.1.0]heptan-2-one (20): To a solution of acetate 18b (85 mg, 0.4 mmol, 1 equiv.) in methanol (3 mL) was added potassium carbonate (113 mg, 0.8 mmol, 2 equiv.), and the mixture was stirred for 10 min at room temp. The reaction was quenched with a saturated ammonium chloride solution. The aqueous phase was extracted with Et₂O. The combined organic layers were washed with brine and dried with MgSO₄. Purification by flash chromatography on silica gel (Et₂O/pentane, 5:95) gave the desired ketone 20 (64 mg, 93% yield) as a single diastereomer, the relative stereochemistry of which was determined by NOESY experiments. **20**: Oil. IR (neat): $\tilde{v} = 3080, 2950, 1680, 1470, 1360$ cm⁻¹. ¹H NMR (200 MHz, CDCl₃): $\delta = 5.60$ (m, 1 H), 4.99–4.93 (m, 2 H), 2.35 (m, 1 H), 2.23–2.01 (m, 2 H), 1.72 (m, 1 H), 1.37– 1.14 (m, 3 H), 1.06 (s, 3 H, CH₃), 1.03 (s, 3 H, CH₃), 1.07–0.89 (m, 2 H) ppm. ¹³C NMR (50 MHz, CDCl₃): δ = 210.3 (C, C=O), 135.7 (CH), 117.1 (CH₂), 42.3 (CH), 36.0 (CH₂), 35.8 (CH₂), 30.2 (CH, CH₃), 29.0 (C), 27.7 (CH₃), 26.5 (CH), 9.7 (CH₂) ppm. C₁₂H₁₈O (178.27): calcd. C 80.84, H 10.18; found C 80.72, H 10.26.

3-But-3-enyl-5,5-dimethylbicyclo[4.1.0]hept-2-en-2-yl Acetate (22) and 3-(2,2-Dimethylbut-3-enyl)bicyclo[4.1.0]hept-2-en-2-yl Acetate (23): Cyclization of precursor 21 (40 mg, 0.17 mmol) with PtCl₂ (2.3 mg, 0.008 mmol), following the GP (80 °C, 1 h) gave, after purification by flash chromatography (Et₂OAc/pentane, 8:92), compound 22 (24 mg, 60%) and compound 23 (12 mg, 30%). 22: IR (neat): $\tilde{v} = 3076, 3002, 2953, 2865, 1753, 1694, 1640 \text{ cm}^{-1}$. ¹H NMR (400 MHz, CDCl₃): $\delta = 5.75$ (m, 1 H), 4.96 (d, J = 18.8 Hz, 1 H), 4.93 (d, J = 9.1 Hz, 1 H), 2.16 (s, 3 H), 2.02 (m, 2 H), 1.93 (m, 2 H), 1.66 (d, J = 16.2 Hz, 1 H), 1.57 (d, J = 16.2 Hz, 1 H), 1.21 (m, 1 H), 1.09 (s, 3 H), 1.06 (m, 1 H), 1.02 (s, 3 H), 0.77 (m, 1 H), 0.66 (m, 1 H) ppm. ¹³C NMR (100 MHz, CDCl₃): $\delta = 169.8$ (C=O), 143.4 (C), 138.8 (CH), 116.5 (C), 114.8 (CH₂, CH=CH₂), 38.5 (CH₂), 31.8 (CH₂), 30.0 (CH₂), 29.9 (CH₃), 28.8 (CH₃), 28.3 (C), 26.8 (CH), 21.2 (CH₃), 13.8 (CH), 10.0 (CH₂) ppm. C₁₅H₂₂O₂ (234.33): calcd. C 76.88, H 9.46; found C 76.85, H 9.54. 23: IR (neat): $\tilde{v} = 3080, 3003, 2959, 2926, 2859, 1755, 1682, 1639 \text{ cm}^{-1}$. ¹H NMR (400 MHz, CDCl₃): $\delta = 5.80$ (dd, J = 17.7, 10.8 Hz, 1 H), 4.96 (m, 2 H), 2.15 (s, 3 H), 2.05-1.78 (m, 4 H), 1.65-1.58 (m, 1 H), 1.42-1.37 (m, 1 H), 1.28-1.24 (m, 2 H), 0.96 (s, 6 H), 0.73 (m, 2 H) ppm. 13 C NMR (100 MHz, CDCl₃): $\delta = 169.3$ (C=O), 148.8 (CH), 145.4 (C), 116.8 (C), 109.7 (CH₂, CH=CH₂), 43.7 (CH₂), 38.2 (C), 27.2 (CH₃), 27.0 (CH₃), 26.6 (CH₂), 21.1 (CH₃), 19.5 (CH₂), 14.4 (CH), 12.4 (CH), 9.2 (CH₂) ppm. C₁₅H₂₂O₂ (234.33): calcd. C 76.88, H 9.46; found C 76.97, H 9.69.

Methyl (2*Z*,4*Z*)-3-Acetyloxy-4-methyl-2,4,7-octatrienoate (29): Cyclization of precursor 24 (166 mg, 0.74 mmol) with PtCl₂ (9.7 mg, 0.036 mmol), following the GP (80 °C, 11 h) afforded, after purification by flash chromatography (EtOAc/hexane, 5:95), product 29 (17 mg, 11 % yield). Oil. ¹H NMR (300 MHz, CDCl₃): δ = 6.03 (t, J = 7.5 Hz, 1 H), 5.80– 5.59 (m, 1 H), 5.71 (s, 1 H), 4.91–4.88 (m, 2 H), 3.58 (s, 3 H), 2.84 (t, J = 7.7 Hz, 2 H), 2.22 (s, 3 H), 1.73 (s, 3 H) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = 168.9, 165.4, 159.6, 135.1, 133.2, 128.9, 116.4, 105.0, 51.9, 33.1, 21.0, 14.1 ppm. MS (70 eV): mlz (%) = 224 (4) [M]⁺, 201 (10), 197 (13), 185 (26), 183 (22), 141 (100), 97 (41), 43 (83). C₁₂H₁₆O₄ (224.25): calcd. C 64.27, H 7.19; found C 64.19, H 7.33.

3-Methyl-5-phenyl-1,2-pentadien-1-yl Acetate (32): Cyclization of precursor **25** (156 mg, 0.72 mmol) with PtCl₂ (10.1 mg, 0.038 mmol), following the GP (5 h at room temp. and then 21 h at 40 °C) afforded, after purification by flash chromatography (EtOAc/hexane, 0.5:99.5), the unreacted starting material **25**

(22 mg) and compound **32** [101.1 mg, 63% (74% taking into account the recovered starting material)]. Oil. IR (neat): $\tilde{v} = 3063$, 3027, 2985, 2922, 1976, 1747, 1603, 1497, 1453, 1368, 1215 cm⁻¹. ¹H NMR (300 MHz, CDCl₃): $\delta = 7.18-7.05$ (m, 6 H), 2.67 (t, J = 8.0 Hz, 2 H), 2.38–2.20 (m, 2 H), 2.04 (s, 3 H), 1.77 (d, J = 2.0 Hz, 3 H) ppm. ¹³C NMR (75 MHz, CDCl₃): $\delta = 189.9$, 169.1, 141.9, 128.8 (2 C), 128.7 (2 C), 126.3, 115.9, 110.6, 37.1, 33.9, 21.4, 21.3 ppm. MS (70 eV): m/z (%) = 216 (1) [M]⁺, 173 (42), 156 (84), 145 (48), 129 (46), 91 (100), 43 (81). $C_{14}H_{16}O_2$ (216.28): calcd. C 77.75, H 7.46; found C 77.65, H 7.24.

1-Methyl-3-(2-methylpropenyl)cyclohex-3-en-1-yl Acetate (35) and 1-Methyl-3-methylene-4-[(methyl)ethylidene]cyclohex-1-yl Acetate (36): Cyclization of precursor 26 (154.0 mg, 0.74 mmol) with PtCl₂ (9.84 mg, 0.037 mmol), following the GP (40 h at 40 °C) afforded, after purification by flash chromatography (EtOAc/hexane, 4:96), a mixture of compounds 35 and 36 (35/36, 4:1 (determined by GLC), 127.8 mg, 83% yield) that we were unable to separate by chromatography. **35** + **36**: IR (neat): $\tilde{v} = 2968$, 2853, 1732, 1651, 1436, 1367, 1236, 1209 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) (major isomer 35): $\delta = 5.53$ (br. s, H-4), 5.47 (br. s, H-7), 2.53 (d, J =17.5 Hz, H-2), 2.28 (d, J = 17.5 Hz, H-2'), 2.20–2.05 (m, H-5, 2 H-6), 1.97 (s, OCOCH₃), 1.76 (br. s, H-5', H-9, H-10), 1.53 (s, H-11) ppm; (minor isomer 36): $\delta = 4.96$ (br. s, H-10), 4.61 (br. d, J =2.1 Hz, H-10'), 3.12 (d, J = 13.5 Hz, H-2), 1.95 (s, OCOCH₃), 1.70 (s, H-5', H-8, H-9), 1.52 (s, H-11) ppm. ¹³C NMR (75 MHz, CDCl₃): δ = (major isomer **35**) 169.6, 133.5, 132.1, 125.7, 122.5, 79.4, 39.9, 30.9, 25.6, 23.3, 22.2, 21.2, 18.5 ppm; (minor isomer **36**) 169.5, 146.2, 128.6, 126.0, 109.7, 81.3, 39.6, 37.7, 28.6, 25.6, 23.3, 20.9, 18.9 ppm. GLC/MS: (70 eV, major isomer 35, retention time: 13.0 min) m/z (%) = 208 (5) [M]⁺, 148 (70), 133 (100), 119 (12), 105 (80), 91 (89), 43 (180); (minor isomer **36**, retention time: 11.0 min) 148 (98), 133 (96), 119 (21), 105 (49), 91 (47), 43 (100). C₁₃H₂₀O₂ (208.30): calcd. C 74.96, H 9.68; found C 74.74, H 9.74.

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